

## Molar Mass of a Volatile Liquid

### Goal

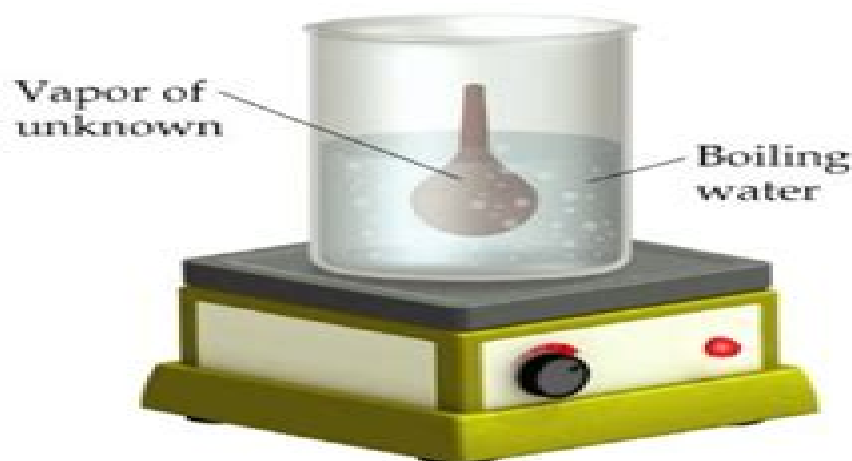
Determine the molar mass of an unknown volatile liquid.

### Background

One of the properties that help characterize a substance is its molar mass. If the substance in question is a volatile liquid, a common method to determine its molar mass is to use the ideal gas law.

$$PV = nRT$$

Because the liquid is volatile, it can easily be converted to a gas. While the substance is in the gas phase, you can measure its volume, pressure, and temperature. You can then use the ideal gas law to calculate the number of moles,  $n$ , of the substance. Finally, you can use the number of moles of the gas to calculate molar mass.



*Figure 1*

### Pre-Laboratory Activity (in lab notebook)

1. Read Ch. 5 in the textbook
2. Identify each variable in the ideal gas Law:  $PV = nRT$  (make sure and specify what units each should be in)
3. A volatile liquid is evaporated in a 250.0 mL Dumas at 100.0 °C and 0.98 atm. At these conditions how many moles of gas would fill the Dumas Bulb?
4. Calculate the Molar Mass of the volatile liquid if the mass of gas that filled the Dumas Bulb was 0.736 g.

**Reminder: All Measurements should be made with the correct number of significant figures and correct units. All calculations results should have the correct number of significant figures and the correct units.**

# Molar Mass Of A Volatile Liquid

**Michael Clugston, Rosalind Flemming**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

## **Molar Mass Of A Volatile Liquid:**

**Determining the Molar Mass of a Volatile Liquid by the Dumas Method** Marcia L. Gillette, John W.

Alcock, 1996-01-01 *Laboratory Manual for Principles of General Chemistry* Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

**Chemical Demonstrations** Bassam Z. Shakhshiri, 1985 Describes and gives instructions for lecture demonstrations covering acids and bases and liquids solutions and colloids A-level Chemistry E. N.

Ramsden, 2000 Each topic is treated from the beginning without assuming prior knowledge Each chapter starts with an opening section covering an application These help students to understand the relevance of the topic they are motivational and they make the text more accessible to the majority of students Concept Maps have been added which together with Summaries throughout aid understanding of main ideas and connections between topics Margin points highlight key points making the text more accessible for learning and revision Checkpoints in each chapter test students understanding and support their private study

**Laboratory Manual for Principles of General Chemistry** J. A. Beran, Mark Lassiter, 2022-08-16 *Laboratory Manual for Principles of General Chemistry* 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section New Understanding Chemistry for Advanced Level Third Edition Ted Lister, Janet Renshaw, 2000 Matches the specifications of the Awarding Bodies AQA NEAB AEB OCR and Edexcel This accessible text includes frequent hints questions and examination questions providing support and facilitating study at home It features photographs and

comprehensive illustrations with 3D chemical structures      *CliffsNotes AP Chemistry 2021 Exam* Angela Woodward Spangenberg, 2020-09-29 *CliffsNotes AP Chemistry 2021 Exam* gives you exactly what you need to score a 5 on the exam concise chapter reviews on every AP Chemistry subject in depth laboratory investigations and full length model practice exams to prepare you for the May 2021 exam Revised to even better reflect the new AP Chemistry exam this test prep guide includes updated content tailored to the May 2021 exam Features of the guide focus on what AP Chemistry test takers need to score high on the exam Reviews of all subject areas In depth coverage of the all important laboratory investigations Two full length model practice AP Chemistry exams Every review chapter includes review questions and answers to pinpoint problem areas      **Chemistry** Sally Solomon, Susan Rutkowski, Charles Boritz, 2008-05-02 *Chemistry An Everyday Approach to Chemical Investigation* is intended to accompany any mainstream general chemistry course and consists of 27 experiments that can be completed using only chemicals found in consumer products The manual is an ideal resource for courses emphasizing green chemistry in which the use of hazardous materials is reduced or eliminated altogether Many of the experiments requiring simple equipment and glassware can be performed at remote sites providing laboratory experience for use with on line or long distance learning courses The advantages of using accessible materials in chemistry laboratory are considerable Students can reinforce lecture discussions while working with familiar materials For instructors assembling the chemicals required for a lab course can be accomplished with limited budgets and without access to a chemical company Problems with safety and waste disposal are significantly reduced      *Ebook: Chemistry: The Molecular Nature of Matter and Change* Silberberg, 2015-01-16 *Ebook Chemistry The Molecular Nature of Matter and Change*      **Handbook of Heavy Oil Properties and Analysis** James G. Speight, 2023-05-12 *Handbook of Heavy Oil Properties and Analysis* Understand the future of oil production with this comprehensive guide Heavy oil also known as viscous oil is oil too viscous to flow normally from wells and reservoirs In recent decades it has become increasingly important as a source of liquid oil for use in industrial processes This places all the greater importance on proper analysis of heavy oil and its properties so that it can be more effectively refined and deployed to meet ever growing energy needs *Handbook of Heavy Oil Properties and Analysis* provides a comprehensive introduction to the analysis of viscous oil and its properties It discusses the full range of tests and analytical procedures by which the behavior and refinability of viscous oil samples can be predicted and connects theoretical knowledge to refinery practice throughout Additionally its incorporation of the latest environmental regulations makes it an invaluable resource Readers will also find Detailed coverage of both physical properties and chemical composition of heavy oil An author more than fifty years of experience in the process industries Discussion of new methods for determining instability and incompatibility This book is a useful reference for scientists and engineers in the oil refining industries chemists and researchers in heavy oil and adjacent industries and government officials and regulators      *General Chemistry* Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie s classic text offers a thorough revision and a

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the exam      **Advanced Chemistry** Michael Clugston, Rosalind Flemming, 2000-06-08 Carefully researched by the authors to bring the subject of chemistry up to date this text provides complete coverage of the new A and AS level core specifications The inclusion of objectives and questions make it suitable for self study      **Problems in Physical Chemistry JEE Main and Advanced Volume 1** Dr. RK Gupta, 2021-04-05 1 The book is prepared for the problem solving in chemistry 2 It is divided into 8 chapters 3 Each chapter is topically divided into quick theory Immediate Test and Knowledge Confirmation Test 4 At the end of the each chapter cumulative exercises for JEE Main Advanced for practice 5 Acid Test for JEE Mains Advance containing all types of questions asked in JEE A common phrase among JEE Aspirants that chemistry is the most scoring subject but the problems asked in JEE Exams are not directly related but they are based on multiple applications Introducing the all new edition of Problem Physical Chemistry JEE Main Advanced Volume 1 which is designed to develop the use of the concepts of chemistry in solving the diversified problems as asked in JEE The book divides the syllabus into 8 chapters and each chapter has been topically divided in quick theory different types of Solved Examination followed by Immediate Test along with the Topicwise short exercises Knowledge Confirmation Test At the end of each chapter there are separate cumulative exercises for JEE Main Advanced Acid Test for JEE Mains Advance are also provided containing all types of questions asked in JEE Detailed and explanatory solutions provided to all the questions for the better understanding TOC Mole concept and Stiochiometry Atomic Structure Stages of Matter 1 Stages of Matter 2 Thermodynamic Thermochemistry Chemical Equilibrium Ionic Equilibrium      **Oswaal NCERT Exemplar (Problems - Solutions) Class 12 Physics, Chemistry and Mathematics (Set of 3 Books) For 2024 Board Exam** Oswaal Editorial Board, 2023-10-28 Description of the product Chapter wise and Topic wise presentation Chapter wise Objectives A sneak peek into the chapter Mind Map A single page snapshot of the entire chapter Revision Notes Concept based study materials Tips Tricks Useful guidelines for attempting each question perfectly Some Commonly Made Errors Most common and unidentified errors are focused Expert Advice Oswaal Expert Advice on how to score more Oswaal QR Codes For Quick Revision on your Mobile Phones and Tablets      **Oswaal NCERT Exemplar (Problems - Solutions) Class 12 Physics, Chemistry and Biology (Set of 3 Books) For 2024 Board Exam** Oswaal Editorial Board, 2023-10-28 Description of the product Chapter wise and Topic wise presentation Chapter wise Objectives A sneak peek into the chapter Mind Map A single page snapshot of the entire chapter Revision Notes Concept based study materials Tips Tricks Useful guidelines for attempting each question perfectly Some Commonly Made Errors Most common and unidentified errors are focused Expert Advice Oswaal Expert Advice on how to score more Oswaal QR Codes For Quick Revision on your Mobile Phones and Tablets      *Instructor's Guide for Introductory Chemistry in the Laboratory* James F. Hall, 1996

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