

35. (a) $0.35744 \text{ mol} \times \frac{6.022 \times 10^{23} \text{ atoms}}{1 \text{ mol}} = 2.152 \times 10^{23} \text{ atoms}$

(b) $2.152 \times 10^{23} \text{ atoms} \times \frac{(14 p^+ + 14 n + 14 e^-)}{1 \text{ atom}} = 9.039 \times 10^{24}$

37. $V_{\text{cube}} = \left(1.25 \text{ in} \times \frac{2.54 \text{ cm}}{1 \text{ in}} \right)^3 = 32.0 \text{ cm}^3$

$32.0 \text{ cm}^3 \times \frac{0.968 \text{ g}}{1 \text{ cm}^3} \times \frac{6.022 \times 10^{23} \text{ atoms}}{22.99 \text{ g}} = 8.12 \times 10^{23} \text{ atoms}$

39. (a) K (b) Cd (c) Al (d) Sb (e) P

41. (a) main-group metal (b) transition metal (c) main-group metal (d) metalloid
(e) nonmetal

43. (a) 6 (b) 4 named, 1 not named (c) 0

45. (a) 13 (b) 2 (c) 17, 18

47. (a) $\text{C}_2\text{H}_7\text{N}$ (b) $\text{C}_3\text{H}_8\text{O}$

49. (a) 14 p^+ , 14 e^- (b) 21 p^+ , 22 e^- (c) 35 p^+ , 34 e^- (d) 70 p^+ , 70 e^-

51.

| | | | | |
|-----------------------------|----|----|----|----|
| ^{19}F 9 | 0 | 9 | 10 | 9 |
| ^{31}P 15 | 0 | 15 | 16 | 15 |
| $^{57}\text{Co}^{2+}$ 27 | +3 | 27 | 30 | 24 |
| $^{32}\text{S}^{2-}$ 16 | -2 | 16 | 16 | 18 |

53. (a) electrolyte (b) nonelectrolyte (c) nonelectrolyte (d) electrolyte

55. (a) CH_4 (b) Cl_4 (c) H_2O_2 (d) NO (e) SiO_2

57. (a) iodine trichloride (b) dinitrogen pentoxide (c) phosphine
(d) carbon tetrabromide (e) sulfur trioxide

59. KCl, K_2S , CaCl_2 , CaS

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Scott C. Dulebohn



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Table of Contents Modern Chemistry Textbook Answers Chapter 8

1. Understanding the eBook Modern Chemistry Textbook Answers Chapter 8
 - The Rise of Digital Reading Modern Chemistry Textbook Answers Chapter 8
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Textbook Answers Chapter 8
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Textbook Answers Chapter 8
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Textbook Answers Chapter 8
 - Personalized Recommendations
 - Modern Chemistry Textbook Answers Chapter 8 User Reviews and Ratings
 - Modern Chemistry Textbook Answers Chapter 8 and Bestseller Lists

5. Accessing Modern Chemistry Textbook Answers Chapter 8 Free and Paid eBooks
 - Modern Chemistry Textbook Answers Chapter 8 Public Domain eBooks
 - Modern Chemistry Textbook Answers Chapter 8 eBook Subscription Services
 - Modern Chemistry Textbook Answers Chapter 8 Budget-Friendly Options
6. Navigating Modern Chemistry Textbook Answers Chapter 8 eBook Formats
 - ePub, PDF, MOBI, and More
 - Modern Chemistry Textbook Answers Chapter 8 Compatibility with Devices
 - Modern Chemistry Textbook Answers Chapter 8 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Chemistry Textbook Answers Chapter 8
 - Highlighting and Note-Taking Modern Chemistry Textbook Answers Chapter 8
 - Interactive Elements Modern Chemistry Textbook Answers Chapter 8
8. Staying Engaged with Modern Chemistry Textbook Answers Chapter 8
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Textbook Answers Chapter 8
9. Balancing eBooks and Physical Books Modern Chemistry Textbook Answers Chapter 8
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Textbook Answers Chapter 8
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Textbook Answers Chapter 8
 - Setting Reading Goals Modern Chemistry Textbook Answers Chapter 8
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Textbook Answers Chapter 8
 - Fact-Checking eBook Content of Modern Chemistry Textbook Answers Chapter 8
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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