CHAPTER 6 REVIEW

Chemical Bonding

SECTION 3

SHORT ANSWER Answer the following questions in the space provided.

- _A_ The notation for sodium chloride, NaCl, stands for one
 (a)formula unit.
 (b)molecule.(d)

 crystal.
 atom.
- _D_ In a crystal of an ionic compound, each cation is surrounded by a number of
 (a)molecules.
 (c) dipoles.
- (a)molecules. (c) dipoles. (b)positive ions. (d) negative ions.
- B Compared with the neutral atoms involved in the formation of an ionic compound, the crystal lattice that results is
- (a)higher in potential energy. (c) equal in potential energy.
- (b)lower in potential energy. (d) unstable.
- B The lattice energy of compound A is greater in magnitude than that of compound B. What can be concluded from this fact?
 (a)Compound A is not an ionic compound.
- (b)It will be more difficult to break the bonds in compound A than those in compound B.
- (c)Compound B has larger crystals than compound A.
- (d)Compound A has larger crystals than compound B.
- B The forces of attraction between molecules in a molecular compound are generally
- (a)stronger than the attractive forces among formula units in ionic bonding.
- (b)weaker than the attractive forces among formula units in ionic bonding.
- (c)approximately equal to the attractive forces among formula units in ionic bonding.
- (d)equal to zero.
- 6.Describe the force that holds two ions together in an ionic bond.

Very strong overall attraction between positive and negative charges

7. What type of energy best represents the strength of an ionic bond?

Lattice energy

8.What types of bonds are present in an ionic compound that contains a polyatomic ion?

Modern Chemistry Review Section 6 Answer Key

David Baud

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The Atmospheric Environment Michael B. Mcelroy, 2021-08-10 This comprehensive introduction to the physics and chemistry of Earth's atmosphere explains the science behind some of the most critical and intensely debated environmental controversies of our day In it one of the world's leading experts on planetary environments presents the background necessary to assess the complex effects of human activity on our atmosphere and climate Unique in its breadth and depth of coverage The Atmospheric Environment includes a survey of Earth's climatic history to provide a context for assessing the changes underway today It is written for and will be of lasting value to a varied audience including not only students but also

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