



Name _____

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Modern Chemistry • CHAPTER 7

HOMEWORK 7-1

(pp. 263–266)

VOCABULARYWrite *true* or *false* for each statement.

1. A chemical formula indicates the relative number of molecules of each kind in a chemical compound.

2. Monoatomic ions are ions formed from a single atom. _____

3. Monoatomic anions of an element are identified by the element's name. _____

4. A binary compound is made of two or more different elements. _____

5. The Stock system is a system for naming chemical elements and their ions. _____

SKILL BUILDER

Explain what each chemical formula tells you about the composition of the compound. The first one is done for you.



The subscript 3 refers to the number of oxygen atoms. There are one atom of sodium and one atom of nitrogen for every three atoms of oxygen.

**STANDARDIZED TEST PREP**

Circle the letter of the best answer.

1. Which statement is true?
- All main-group elements are able to form ions.
 - The total numbers of positive charges and negative charges must be equal in a binary compound.
 - The nonmetals of groups 15, 16, and 17 lose electrons to form cations.
 - Elements in the *d* block form ions of only one charge.
2. Why do many main-group elements lose or gain electrons to form ions?
- The elements are unstable and therefore lose electrons.
 - Their tendency is to maximize potential energy.
 - The elements form covalent bonds, sharing their electrons with other atoms.
 - Their tendency is to form a complete outermost octet in a noble-gas configuration.

Modern Chemistry

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Modern Science and the Book of Genesis James William Skehan, National Science Teachers Association, 1986 Based on the premise that knowledge of evolutionary theory is essential for understanding the natural world this document was designed to assist science teachers and others as they consider the issues that influence the teaching of evolution The position is taken that there is no conflict between data and sound theories based on science and religious beliefs based on the Bible Information and perspectives are presented under the topic headings of 1 The Genesis of Genesis 2 Early Science Interprets Genesis 3 New Data 4 Creationism versus Science and 5 Two Kinds of Knowledge References are listed and the National Science Teacher Association s position statement on the Inclusion of Nonscience Tenets in Science Instruction is included ML **Fundamentals of Chemistry: A Modern Introduction** Frank Brescia, 2012-12-02 Fundamentals of

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