

## Chapter Test B

### Chapter: Chemical Equations and Reactions

Write a correct name (including state) for the react or product described. Completed tests must include correct answers for all questions.

- \_\_\_\_\_ 1. The product of a slightly soluble substance in a double displacement reaction is the formation of a  
a. gas  
**b. precipitate**  
c. insoluble substance  
d. solution
- \_\_\_\_\_ 2. An ionic compound equation is represented by which the  
**a. products**  
b. reactants  
c. formation of the products  
d. transfer of products
- \_\_\_\_\_ 3. An ionic equation shows the activity of  $\text{aq}$  ions in the solution  
a. water  
**b. dissolved in water**  
c. reactants  
d. reactants
- \_\_\_\_\_ 4. The following list of substances are used to produce a gas  
a. sodium hydroxide, water, aluminum, and hydrochloric acid  
b. hydrogen, aluminum  
c. sodium hydroxide, water  
**d. calcium hydroxide,  $\text{aq}$ , acid, aluminum, and hydrochloric acid**
- \_\_\_\_\_ 5. The following are chemical equations  
**a.  $\text{NaOH(aq)} + \text{HCl(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$**   
b.  $\text{NaOH(aq)} + \text{HCl(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$   
c.  $\text{NaOH(aq)} + \text{HCl(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$   
d.  $\text{NaOH(aq)} + \text{HCl(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
- \_\_\_\_\_ 6. It is known that  $\text{NaOH}$  is a base. If  $\text{NaOH}$  is a base, then  
a.  $\text{NaOH}$  is a base  
**b.  $\text{NaOH}$  is a base**  
c.  $\text{NaOH}$  is a base  
d.  $\text{NaOH}$  is a base

# Modern Chemistry Chapter 8 Test B Answers

**Jianjun Gao**



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