

1. What are unique properties of water? List them and explain each in a single sentence.

Cohesion- sticking of one substance to another

Solvent - anything that has ionic bonds or polar covalent will dissolve in water easily.

High specific heat- amount of heat needed to change temperature

High heat evaporation- must absorb additional heat to change from liquid to gas

Lower density of ice- ice's ability to float

OK by Dr. Lee

2. Why does water have those unique properties? What is water made of and what types of chemical bonds does it have?

One oxygen atom covalently bonded through polar bond to two hydrogen atoms. It acts so uniquely because its structure gives it a partial polar charge, making it easy to break down molecules that enter water and also easy to bond with each other.

OK by Dr. Lee

3. What is an element?

Element is made out of atoms of one type. It is a pure substance that cannot be broken down by chemical means into other substances.

OK by Dr. Lee

4. What is an atom made of?

Electrons, neutrons, and protons

OK by Dr. Lee

5. This is Cesium. Based on figure below, answer questions a-j. When answering these questions assume that 1st shell holds 2 electrons and ALL next shells can hold only 8 electrons (do not rely on information from internet in this case)

Modern Biology Chapter2 Review Answers

Yicheng Fang



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