Name Class

Assessment)

Chapter Test A

Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- Which of the properties listed below is not characteristic of an acid?
 - a. a sour taste
 - b. a slippery feel
 - c. the ability to conduct an electric current
 - d. reactivity with metals
 - 2. Which is an example of a weak acid?
 - a. HCI
 - **b.** H₂SO₄
 - c. HaPO₄
 - d. HNO_a
- _____ 3. Acetic acid is found in
 - a. vinegar.
 - b. the stomach.
 - c. antacids.
 - d. oranges.
- _____ 4. A characteristic of an Arrhenius base is that it
 - a. is an electrolyte.
 - b. tastes bitter.
 - c. dissociates to form OH ons.
 - d. All of the above
- 5. Which of the following is a Brønsted-Lowry base?
 - a. HCl
 - b. HCO_a
 - c. H₀O *
 - d. H_0PO_4

6. In the reaction represented by the equation

$$H_2C_2O_4(aq) + CH_3NH_2(aq) \Leftrightarrow HC_2O_4^-(aq) + CH_3NH_2^+(aq),$$

which of these is a conjugate acid-base pair?

- a. $H_2C_2O_4(aq)$ and $CH_3NH_2(aq)$
- b. H₂C₂O₄(aq) and CH₃NH^{*}₃(aq)
- c. CH₃NH₂(aq) and CH₃NH'₃(aq)
- d. HC2O4(aq) and CH2NH2(aq)

Disable:

Modern Chemistry Chapter 3 Test

Lauren Gardner

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Modern Chemistry Chapter 3 Test Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "Modern Chemistry Chapter 3 Test," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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