

# HYDROCARBONS



## Hydrocarbon Compounds

Hydrocarbon compounds are compounds that consist only of carbon and hydrogen atoms. Hydrocarbons can be divided into 2, namely:

### a. Saturated Hydrocarbons: single bonds

including saturated hydrocarbons :

- Alkane : open chain ( $C_nH_{2n+2}$ )
- Cycloalkane : closed chain ( $C_nH_{2n}$ )

### b. Unsaturated Hydrocarbons: double bonds

- Alkene : double bond ( $C_nH_{2n}$ )
- Alkyne : triple bond ( $C_nH_{2n-2}$ )
- Alkadiene : two double bonds ( $C_nH_{2n-2}$ )



## Nomenclature of Alkanes

The nomenclature of alkanes is the basis for the nomenclature of other carbon compounds. Therefore the nomenclature of alkanes must be thoroughly understood.

Here are the names of alkanes from  $C_1$ - $C_{10}$  :

Formula	Name	Formula	Name
$CH_4$	methane	$C_6H_{14}$	hexane
$C_2H_6$	ethane	$C_7H_{16}$	heptane
$C_3H_8$	propane	$C_8H_{18}$	octane
$C_4H_{10}$	butane	$C_9H_{20}$	nonane
$C_5H_{12}$	pentane	$C_{10}H_{22}$	decane

Alkene names :

$C_2H_4$  = ethylene

$C_3H_6$  = propene

# Hydrocarbon Study Guide

**RJ Alexander**



## **Hydrocarbon Study Guide:**

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## **Table of Contents Hydrocarbon Study Guide**

1. Understanding the eBook Hydrocarbon Study Guide
  - The Rise of Digital Reading Hydrocarbon Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Hydrocarbon Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hydrocarbon Study Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Hydrocarbon Study Guide
  - Personalized Recommendations
  - Hydrocarbon Study Guide User Reviews and Ratings
  - Hydrocarbon Study Guide and Bestseller Lists
5. Accessing Hydrocarbon Study Guide Free and Paid eBooks
  - Hydrocarbon Study Guide Public Domain eBooks
  - Hydrocarbon Study Guide eBook Subscription Services
  - Hydrocarbon Study Guide Budget-Friendly Options
6. Navigating Hydrocarbon Study Guide eBook Formats

- ePub, PDF, MOBI, and More
- Hydrocarbon Study Guide Compatibility with Devices
- Hydrocarbon Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Hydrocarbon Study Guide
  - Highlighting and Note-Taking Hydrocarbon Study Guide
  - Interactive Elements Hydrocarbon Study Guide
- 8. Staying Engaged with Hydrocarbon Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hydrocarbon Study Guide
- 9. Balancing eBooks and Physical Books Hydrocarbon Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hydrocarbon Study Guide
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Hydrocarbon Study Guide
  - Setting Reading Goals Hydrocarbon Study Guide
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hydrocarbon Study Guide
  - Fact-Checking eBook Content of Hydrocarbon Study Guide
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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