

Chapter 9 Carbon Chemistry

Section 9.2 Substituted Hydrocarbons**(pages 272–274)**

This section discusses organic compounds that contain atoms of elements other than carbon and hydrogen. It also explains the relationship between the properties of organic compounds and functional groups.

Reading Strategy (page 272)

Monitoring Your Understanding As you read, complete the table by connecting each functional group with the type of compound that contains the functional group. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Connecting Functional Groups to Types of Compounds	
Functional Group	Type of Compound
–OH	
–COOH	
–NH ₂	

- Name the two main products when methane and chlorine react.
 - _____
 - _____
- To which environmental problem have researchers connected halocarbons containing chlorine and fluorine? _____

- Describe a substituted hydrocarbon. _____

- Is the following sentence true or false? The functional group in a substituted hydrocarbon determines the properties of the compound. _____

Alcohols (page 273)

- Methanol and ethanol are two examples of a class of organic compounds called _____.
- The functional group in an alcohol is represented as –OH and is called a(n) _____ group.
- Identify two ways a halocarbon can be produced.
 - _____
 - _____

Hydrocarbon Compounds Section Review Answer Key

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