

Name \_\_\_\_\_ Class \_\_\_\_\_ Date \_\_\_\_\_

## Assessment

## Chapter Test B

## Atomic Physics

## MULTIPLE CHOICE

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. What is the frequency of a photon with an energy of  $1.00 \times 10^{-19} \text{ J}$ ?  
( $h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$ )  
a.  $1.00 \times 10^{14} \text{ Hz}$  c.  $3.00 \times 10^{12} \text{ Hz}$   
b.  $2.00 \times 10^{14} \text{ Hz}$  d.  $4.00 \times 10^{12} \text{ Hz}$
- \_\_\_\_\_ 2. Light with an energy equal to three times the work function of a given metal causes the metal to eject photoelectrons. What is the ratio of the maximum photoelectron kinetic energy to the work function?  
a. 1 : 1 c. 3 : 1  
b. 2 : 1 d. 4 : 1
- \_\_\_\_\_ 3. A monochromatic light beam with a quantum energy value of 3.0 eV is incident upon a photocell. The work function of the photocell is 1.6 eV. What is the maximum kinetic energy of the ejected electrons?  
a. 4.6 eV c. 1.4 eV  
b. 4.8 eV d. 2.4 eV
- \_\_\_\_\_ 4. What causes the bright lines in the emission spectrum of an element to occur?  
a. Photons are absorbed when electrons jump from a higher-energy to a lower-energy state.  
b. Photons are emitted when electrons jump from a higher-energy to a lower-energy state.  
c. Photons are absorbed when electrons jump from a lower-energy to a higher-energy state.  
d. Photons are emitted when electrons jump from a lower-energy to a higher-energy state.
- \_\_\_\_\_ 5. What causes the dark lines in the absorption spectrum of an element to occur?  
a. Photons are absorbed when electrons jump from a higher-energy to a lower-energy state.  
b. Photons are emitted when electrons jump from a higher-energy to a lower-energy state.  
c. Photons are absorbed when electrons jump from a lower-energy to a higher-energy state.  
d. Photons are emitted when electrons jump from a lower-energy to a higher-energy state.

# Holt Physical Science Chapter 14 Chemical Reactions

## Chapter Test

**Rachel Sandford**



**Holt Physical Science Chapter 14 Chemical Reactions Chapter Test:**

## Adopting the Melody of Appearance: An Emotional Symphony within **Holt Physical Science Chapter 14 Chemical Reactions Chapter Test**

In a world eaten by monitors and the ceaseless chatter of fast interaction, the melodic beauty and mental symphony produced by the published term often diminish in to the background, eclipsed by the constant noise and distractions that permeate our lives. Nevertheless, set within the pages of **Holt Physical Science Chapter 14 Chemical Reactions Chapter Test** an enchanting literary prize filled with raw feelings, lies an immersive symphony waiting to be embraced. Constructed by a masterful composer of language, this charming masterpiece conducts viewers on a mental journey, well unraveling the hidden melodies and profound affect resonating within each carefully constructed phrase. Within the depths with this poignant analysis, we shall discover the book is main harmonies, analyze their enthralling publishing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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