

## 042723 PHYSICAL SCIENCE CH 17 TEST REVIEW

DAVIS

NAME: \_\_\_\_\_

04/27/23

1. What are the 4 general characteristics of metals?

Malleable, ductile, shiny, good conductors

2. Define ductile.

able to be drawn out into a thin wire.

3. Define malleable.

able to be hammered or pressed permanently out of shape without breaking or cracking.

4. What action do metals prefer to take with their valence electrons during bonding w/ nonmetals?

Give them away

5. What happens to their charge when they do so?

Stays positive

6. What do we call atoms with this charge?

Cations

7. What do we call the type of bonding that occurs when metals and nonmetals bond?

Ionic bond

8. What type of bond forms between metals and other metallic atom?

Metallic bond

9. The electrons in metallic bonds exist in a "\_\_\_cloud\_\_\_" on the surface.

10. Which important characteristic of metals does this permit?

luster

11. What group contains the Alkali Metals?

IA

12. What is the name of group IIA?

Alkali Earth Metals

# Holt Physical Science Chapter17 Sectionreview Answers

**Ralph Thomas Myers**



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