

CHAPTER 14 STUDY GUIDE

Mixtures and Solutions

Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

- | | |
|--------------|---|
| <u>false</u> | 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed. |
| <u>true</u> | 2. The most abundant substance in a colloid is the dispersion medium. |
| <u>false</u> | 3. A colloid can be separated by filtration. |
| <u>true</u> | 4. A solid emulsion consists of a liquid dispersed in a solid. |
| <u>true</u> | 5. Whipped cream is an example of a foam. |
| <u>false</u> | 6. In an aerosol, the dispersing medium is a liquid. |
| <u>true</u> | 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles. |
| <u>true</u> | 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces. |
| <u>false</u> | 9. Stirring an electrolyte into a colloid stabilizes the colloid. |
| <u>true</u> | 10. Colloids demonstrate the Tyndall effect. |

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter		☒	☒
12. Between 1 nm and 1000 nm in diameter	☒	☒	☒
13. More than 1000 nm in diameter	☒	☒	
14. Settle out if undisturbed			
15. Pass through standard filter paper	☒		
16. Lower vapor pressure			☒
17. Scatter light	☒		

Heterogeneous Mixtures Study Guide Teacher Eddition

Clyde Metz



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Table of Contents Heterogeneous Mixtures Study Guide Teacher Edition

1. Understanding the eBook Heterogeneous Mixtures Study Guide Teacher Edition
 - The Rise of Digital Reading Heterogeneous Mixtures Study Guide Teacher Edition
 - Advantages of eBooks Over Traditional Books
2. Identifying Heterogeneous Mixtures Study Guide Teacher Edition
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Heterogeneous Mixtures Study Guide Teacher Edition
 - User-Friendly Interface
4. Exploring eBook Recommendations from Heterogeneous Mixtures Study Guide Teacher Edition
 - Personalized Recommendations
 - Heterogeneous Mixtures Study Guide Teacher Edition User Reviews and Ratings
 - Heterogeneous Mixtures Study Guide Teacher Edition and Bestseller Lists
5. Accessing Heterogeneous Mixtures Study Guide Teacher Edition Free and Paid eBooks
 - Heterogeneous Mixtures Study Guide Teacher Edition Public Domain eBooks
 - Heterogeneous Mixtures Study Guide Teacher Edition eBook Subscription Services
 - Heterogeneous Mixtures Study Guide Teacher Edition Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Heterogeneous Mixtures Study Guide Teacher Edition Compatibility with Devices
 - Heterogeneous Mixtures Study Guide Teacher Edition Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Heterogeneous Mixtures Study Guide Teacher Edition
 - Highlighting and Note-Taking Heterogeneous Mixtures Study Guide Teacher Edition
 - Interactive Elements Heterogeneous Mixtures Study Guide Teacher Edition
8. Staying Engaged with Heterogeneous Mixtures Study Guide Teacher Edition
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Heterogeneous Mixtures Study Guide Teacher Edition
9. Balancing eBooks and Physical Books Heterogeneous Mixtures Study Guide Teacher Edition
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Heterogeneous Mixtures Study Guide Teacher Edition
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Heterogeneous Mixtures Study Guide Teacher Edition
 - Setting Reading Goals Heterogeneous Mixtures Study Guide Teacher Edition
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Heterogeneous Mixtures Study Guide Teacher Edition
 - Fact-Checking eBook Content of Heterogeneous Mixtures Study Guide Teacher Edition
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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