General Instructions:

Read the following instructions very carefully and strictly follow them:

- (i) This question paper comprises 15 questions. All questions are compulsory.
- (ii) This question paper is divided into three sections A, B and C.
- Section A Questions No. 1 to 7 are short answer type questions. Each question carries 2 marks.
- Section B Questions No. 8 to 13 are also short answer type questions. Each question carries 3 marks.
- Section C Questions No. 14 and 15 are case-based questions. Each question carries 4 marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

SECTION A

- 1. (a) Write the molecular formula of the following carbon compounds:
 - (i) Methane
 - (ii) Propane
 - (b) Carbon compounds have low melting and boiling points. Why?
- The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.
 - (a) Write the group numbers of these elements in the Modern Periodic Table.
 - (b) Write the molecular formula of the compound formed when X and Y combine with each other.
 - (c) Which of the two is electropositive?
- 3. (a) Which of the following flowers will have higher possibility of self-pollination?

Mustard, Papaya, Watermelon, Hibiscus

- (b) List the two reproductive parts of a bisexual flower.
- 4. Which one of the two multicellular organisms Spirogyra and Planaria reproduces by regeneration and why? Give an example of any other organism which can also reproduce by the same process.

31/1/1





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Daniela Niemeyer

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