

## Exam I Study Guide

- **Physical Changes vs. Chemical Changes (Can you undo it?)**
  - Physical changes result in changes of appearance, but not composition.
    - Melting, evaporation, freezing, etc.
  - Chemical changes result in changes of composition of matter. (Chemical reaction)
    - Burning, mixing, dissolving, etc.
- **Physical Properties vs. Chemical Properties**
  - Physical properties can be described by observing the matter without a change in the composition.
    - Color, odor, density, temperature, etc.
  - Chemical properties can only be detected by altering the composition.
    - Flammability, oxidization, etc.
- **Atoms vs. Molecules**
  - Atoms are the smallest unit of a given element that carry the properties of that element.
  - Molecules are composed of two or more atoms.
- **Elements vs. Compounds**
  - Elements occur in nature and cannot be broken down to simpler particles through ordinary means.
  - Compounds are the combination of two or more elements.
- **Accuracy vs. Precision**
  - Accuracy refers to the agreement of a measurement with a known or accepted value. (Bullseye)
    - Correctness
  - Precision refers to the agreement of a measurement with other measurements in a given set of measurements. (Missing the bullseye but landing all darts in one place)
    - Repeatability or reproducibility
- **Rules Concerning Significant Figures**
  - All nonzero digits ARE significant.
  - Zeros between nonzero digits ARE significant.
  - Zeros at the beginning of a number are NOT significant; they only indicate the position of the decimal point.
  - Zeros at the end of a number to the right of a decimal point ARE significant.
  - Zeros at the end of a number with no decimal point are ASSUMED to be NOT significant, UNLESS we indicate their significance by placing a line over the last zero that is significant.

# General Chemistry Standardized Final Exam Study Guide

**L Towner**



## **General Chemistry Standardized Final Exam Study Guide:**

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Aleksij Glažar, 2014-01-14 This volume offers a critical examination of a variety of conceptual approaches to teaching and learning chemistry in the school classroom Presenting up to date research and theory and featuring contributions by respected academics on several continents it explores ways of making knowledge meaningful and relevant to students as well as strategies for effectively communicating the core concepts essential for developing a robust understanding of the subject Structured in three sections the contents deal first with teaching and learning chemistry discussing general issues and pedagogical strategies using macro sub micro and symbolic representations of chemical concepts Researchers also describe new and productive teaching strategies The second section examines specific approaches that foster learning with understanding focusing on techniques such as cooperative learning presentations laboratory activities multimedia simulations and role playing in forensic chemistry classes The final part of the book details learner centered active chemistry learning methods active computer aided learning and trainee chemistry teachers use of student centered learning during their pre service education Comprehensive and highly relevant this new publication makes a significant contribution to the continuing task of making chemistry classes engaging and effective

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Chemistry DeMYSTiFieD, Second Edition Linda D. Williams, 2011-05-13 A PROVEN formula for mastering CHEMISTRY Trying to understand chemistry but feel like the information s just not bonding with your brain Here s your solution Chemistry Demystified Second Edition helps you grasp both fundamental and complex concepts with ease Written in a step by step format this practical guide first covers atomic theory elements symbols and the Periodic Table of the Elements The book then delves into solids liquids gases solutions orbitals chemical bonds acids and bases Electrochemistry thermodynamics biochemistry and organic environmental and nuclear chemistry are discussed In depth examples detailed illustrations and worked out problems make it easy to understand the material and end of chapter quizzes and a final exam help reinforce learning It s a no brainer You ll learn about Molecular and structural formulas Metallurgy Gas laws Molar mass Molecular orbital theory Covalent and ionic bonds Oxidation reduction The laws of thermodynamics Organic reactions Biological and environmental markers Simple enough for a beginner but challenging enough for an advanced student Chemistry Demystified Second Edition helps you master this fascinating subject MCAT General Chemistry Review 2018-2019 Kaplan Test Prep, 2017-07-04 Kaplan s MCAT Complete 7 Book Set Subject Review has all the information and strategies you need to score higher on the MCAT These books feature more practice than any other guide plus targeted strategy review opportunities for self analysis and thorough information on all of the critical thinking skills necessary for MCAT success from the creators of the 1 MCAT prep course From publisher s description **Teaching Innovation in University Education: Case Studies and Main Practices** Saura, Jose Ramon, 2022-06-17 In the last decade the

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