Orbitals and Bonding Test #12 SCH4U Name:

Multiple Choice: Circle the most correct answer. (15K)

- 1. What is the electron configuration of potassium?
- A) 1s22s22p23s23p24s1

C) 1s22s23s23p43d1

B) 1s²2s²2ρ¹⁰3s²3ρ³

- D) 1s22s2p63s23p64s1
- 2. Which electron configuration represents a violation of the Pauli exclusion principle?

1s 2s 2p

1s 2s 2p

D) ↑↓ ↑↓ ↑ ↑

1s 2s 2p 1 1 1 1 1 1

3. Which one of the following sets of quantum numbers can correctly represent a 3p orbital?

A) B) C) D) E) n=3 n=1 n=3 n=3 n=3 l=1 l=3 l=2 l=1 l=0 $m_1=2$ $m_2=3$ $m_3=1$ $m_4=-1$ $m_5=1$

- 4. How many electrons in an atom can have n = 4, l = 2?
- A) 6 B) 14 C) 10 D) 2 E) None of these

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Kaplan Test Prep

General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test:

This book delves into General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test. General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test
 - Chapter 2: Essential Elements of General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test
 - o Chapter 3: General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test in Everyday Life
 - Chapter 4: General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test in Specific Contexts
 - ∘ Chapter 5: Conclusion
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 This chapter will explore what General Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test is, why General
 Chemistry Ii Chapter Covalent Bonding Orbitals Practice Test is vital, and how to effectively learn about General Chemistry Ii
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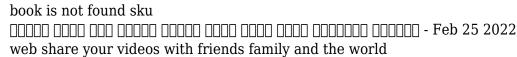
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