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1. Which among the following is not a redox reaction?
 - A. Hydrogen Burnt
 - B. Iron(II)oxide Burnt
 - C. Disproportionation Reaction
 - D. Disproportionation Reaction
2. In which of the following would dipole-dipole forces exist?
 - A. Methane, nonpolar-molecules
 - B. Methanol, H-bond-molecules
 - C. Ethanol is miscible with CCl_4
 - D. Ethanol H-bond-molecules in water
3. The vapour pressure of alcohol at 40 °C is 120.2 kPa. What would be the vapour pressure at 30 °C?
 - A. 40° 50 kPa
 - B. Greater than 40° 50 kPa
 - C. Less than 40° 50 kPa
 - D. Cannot be determined
4. Which of the following solid-molecules, which is hydrogen-bonded will show maximum solubility?
 - A. all the groups of phenol-molecules
 - B. all-methanol
 - C. It would have same since its with hydrogen-bond
 - D. It depends on the polarity of force
5. Which of the following would have the lowest melting point?
 - A. NaBr
 - B. CaBr_2
 - C. NaCl
 - D. graphite
6. Which of the following substance would be a good conductor of H^+ ?
 - A. Methanol
 - B. water
 - C. Ethanol-50%
 - D. None of these
7. Which of the following describes Henry's Law?
 - A. Relationship between pressure and the solubility of gases.
 - B. Relationship between temperature and the solubility of gases.
 - C. Relationship between pressure and the solubility of solids.
 - D. Relationship between temperature and the solubility of solids.
8. How much of propane is 0.100 kg solution of MgCl_2 . how much dissolved the solvent for 1.0mole of MgCl_2 ?
 - A. 1.00 mol
 - B. 1.00 L
 - C. 1.00 mol %
 - D. enough water to make 1.0L of solution
9. Water's vapour-pressure at 20 °C is 2.33 kPa. The temperature, where total amount of water, which is 0.01 kg, has 0.01 kg of H_2O . Determine that temperature that vapour-pressure:
 - A. 0.20 kPa
 - B. 1.00 kPa
 - C. 0.01 kPa
 - D. 0.00 kPa
10. Gas-molecules can be many different molecules. Given pressure, temperature (10.0 kPa) and volume (0.01 m³) early molecule with 1.00-mol of it, molecules, respectively, compare for the number of molecules.

General Chemistry Diagnostic Test

SJ Ball



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