

General Chemistry 10th

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CHAPTER 1

Chemistry and Measurement

■ SOLUTIONS TO EXERCISES

Note on significant figures: If the final answer to a solution needs to be rounded off, it is given first with one nonsignificant figure, and the last significant figure is underlined. The final answer is then rounded to the correct number of significant figures. In multistep problems, intermediate answers are given with at least one nonsignificant figure; however, only the final answer has been rounded off.

- 1.1. From the law of conservation of mass,

$$\text{Mass of wood} + \text{mass of air} = \text{mass of ash} + \text{mass of gases}$$

Substituting, you obtain

$$1.85 \text{ grams} + 9.45 \text{ grams} = 0.28 \text{ grams} + \text{mass of gases}$$

or,

$$\text{Mass of gases} = (1.85 + 9.45 - 0.28) \text{ grams} = 11.02 \text{ grams}$$

Thus, the mass of gases in the vessel at the end of the experiment is 11.02 grams.

- 1.2. Physical properties: soft, silvery-colored metal; melts at 64°C.
Chemical properties: reacts vigorously with water, with oxygen, and with chlorine.

- 1.3. a. The factor 9.1 has the fewest significant figures, so the answer should be reported to two significant figures.

$$\frac{5.61 \times 7.891}{9.1} = 4.\underline{86} = 4.9$$

- b. The number with the least number of decimal places is 8.91. Therefore, round the answer to two decimal places.

$$8.91 - 6.435 = 2.\underline{475} = 2.48$$

- c. The number with the least number of decimal places is 6.81. Therefore, round the answer to two decimal places.

$$6.81 - 6.730 = 0.\underline{080} = 0.08$$

- d. You first do the subtraction within parentheses. In this step, the number with the least number of decimal places is 6.81, so the result of the subtraction has two decimal places. The least significant figure for this step is underlined.

$$38.91 \times (6.81 - 6.730) = 38.91 \times 0.\underline{080}$$

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Table of Contents General Chemistry 10th Edition 15answer

1. Understanding the eBook General Chemistry 10th Edition 15answer
 - The Rise of Digital Reading General Chemistry 10th Edition 15answer
 - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry 10th Edition 15answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Chemistry 10th Edition 15answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry 10th Edition 15answer
 - Personalized Recommendations

- General Chemistry 10th Edition 15answer User Reviews and Ratings
- General Chemistry 10th Edition 15answer and Bestseller Lists
- 5. Accessing General Chemistry 10th Edition 15answer Free and Paid eBooks
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 - General Chemistry 10th Edition 15answer eBook Subscription Services
 - General Chemistry 10th Edition 15answer Budget-Friendly Options
- 6. Navigating General Chemistry 10th Edition 15answer eBook Formats
 - ePub, PDF, MOBI, and More
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 - General Chemistry 10th Edition 15answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Chemistry 10th Edition 15answer
 - Highlighting and Note-Taking General Chemistry 10th Edition 15answer
 - Interactive Elements General Chemistry 10th Edition 15answer
- 8. Staying Engaged with General Chemistry 10th Edition 15answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Chemistry 10th Edition 15answer
- 9. Balancing eBooks and Physical Books General Chemistry 10th Edition 15answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Chemistry 10th Edition 15answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine General Chemistry 10th Edition 15answer
 - Setting Reading Goals General Chemistry 10th Edition 15answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of General Chemistry 10th Edition 15answer
 - Fact-Checking eBook Content of General Chemistry 10th Edition 15answer

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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