(1.6-3)

From appendix A.4, $C_{\pm} = 4.02 \text{ kJ/kg·K at } 32.8 °C$ assume $C_{\pm}m = 4.02$ Note: $(t-10)^{\circ}C = (t-10)^{\circ}K$ Q = 121300 kJ = 454 kg (4.02 kJ)(t-10) KSolving, (t-10) = 66.5 K = 66.5 °C t = 76.5 °C

(1.6-4)

- (a) A.2-9 P=30 baia liquid H20 $\Delta H = H_{240*F} H_{40*F}$ = 208.44 8.02 = 200.42 Gtallerm
- (b) P = 24.97 pera H₂40.F (vageor) = 1160.7 H_{40.F} (lequid) = 8.02 DH = 1152.76te/lbm
- (c) P = 1.0 atm aba

 H212°F (Vopen) = 1150.5

 H400F (liquid) = 28.1

 \[\text{DH} = -1122.4 Gta / \text{Et}_m \]
- (d) P = 1.0 atm aba

 H = 12°F (vapor) = 1150.5

 H = 12°F (lequid) = 180.2

 DH = -970.3 ttc /ttm

 (-2256.9 k 5/kg)

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Jin-Ying Zhang

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