

Dynamic Equilibrium

At what point is a reversible reaction “completed?”

Why?

Most of the reactions we have studied year have been “forward” reactions—once the reactant has been changed into the product it stays that way. The reaction “stops” when the limiting reactant runs out.

However, in a reversible reaction (one that takes place in both directions), the product can turn back into reactant in a continuous cycle that never stops. So...

When can we say the reaction is “finished?”

Can we predict how much reactant and how much product will be present when the reaction is “done?”

Model 1 – A Reversible Reaction



1. What is the significance of the double arrow in Model 1?
2. Imagine the reaction in Model 1 starts with 100 molecules of A and zero molecules of B in a 1 liter container. Explain why the concentration of reactant A will never reach zero.
3. Imagine the reaction in Model 1 starts with zero molecules of A and 100 molecules of B in a 1 liter container. Will the concentration of product B increase, decrease, or remain the same as the reaction proceeds? Explain.

Equilibrium A Reversible Reaction Pogil Answers

Michael Rajnik



Equilibrium A Reversible Reaction Pogil Answers:

Unveiling the Magic of Words: A Review of "**Equilibrium A Reversible Reaction Pogil Answers**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Equilibrium A Reversible Reaction Pogil Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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