

Calorimetry Lab Report

Lab Members: Dylan Craig, and Brayden

Objective: The objective of this lab is to find out if a fishing weight was actually made of lead, to investigate the first law of thermodynamics and Calorimetry, and to design an experiment by first creating an analytical model as a guide.

Introduction: By combining the first law of thermodynamics with Calorimetry, one can determine if a fishing weight that claims to be lead, is actually made of lead or not. The first law of thermodynamics states that you can change the internal energy of a system by doing work or applying heat to a system.

$$\Delta E_{\text{system}} = W_{\text{on}} + Q$$

This can help determine the heat capacity of the lead weight. The heat [Q] is a physical quantity that characterizes the transfer of energy in the form of heat from an environment to a system (pg. 1, Calorimetry).

$$Q = mc \Delta T$$

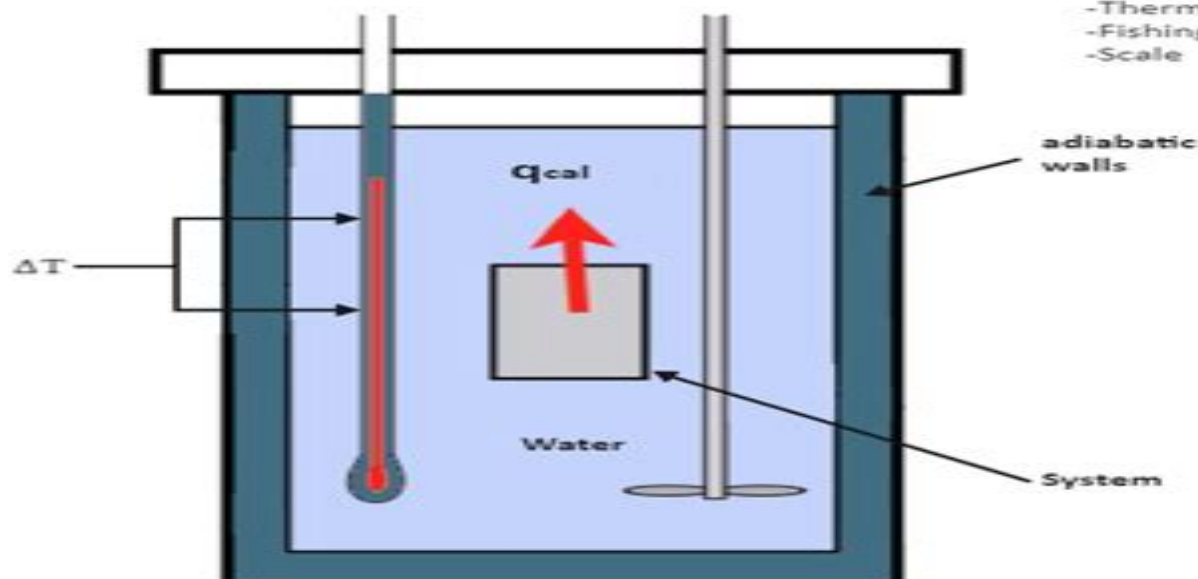
It is important to note that when one is analyzing two objects that are exchanging heat, the conservation of energy is:

$$Q_{\text{hot to cold}} + Q_{\text{cold to hot}} = 0$$

Experimental Setup:

Equipment Used:

- Calorimeter
- Water
- Thermometer
- Fishing Weight
- Scale



Enthalpy Calorimeter Lab Report

Ensheng Dong

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Abstracts of AF Materials Laboratory Reports Air Force Materials Laboratory (U.S.),1968-07 Chemistry in the Laboratory James M. Postma,Julian L. Robert,J. Leland Hollenberg,2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations Chemical Energetics, Equilibria and Functional Group Organic Chemistry I - Laboratory Mr. Rohit Manglik,2024-03-03 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels **Summaries of Reports of the Electrotechnical Laboratory** Denki Shikenjo (Japan),1976 *Scientific and Technical Aerospace Reports* ,1991 *A House Built on Sand* Noretta Koertge,1998-08-27 Cultural critics say that science is politics by other means arguing that the results of scientific inquiry are profoundly shaped by the ideological agendas of powerful elites They base their claims on historical case studies purporting to show the systematic intrusion of sexist racist capitalist colonialist and or professional interests into the very content of science Physicist Alan Sokal recently poked fun at these claims by foisting a sly parody of the genre on the unwitting editors of the cultural studies journal Social Text touching off a still unabated torrent of editorials articles and heated classroom and Internet discussion This hard hitting collection picks up where Sokal left off The essayists offer crisp and detailed critiques of case studies offered by the cultural critics as evidence that scientific results tell us more about social context than they do about the natural world Pulling no punches they identify numerous crude factual blunders e g that Newton never performed any experiments and egregious errors of omission such as the attempt to explain the slow development of fluid dynamics solely in terms of gender bias Where there are positive aspects of a flawed account or something to be learned from it they do not hesitate to say so Their target is shoddy scholarship Comprising new essays by distinguished scholars of history philosophy and science including Sokal himself this book raises a lively debate to a new level of seriousness **Nuclear Science Abstracts** ,1963-10 CliffsNotes AP Chemistry Angela Woodward Spangenberg,2016-01-12 Test prep for the AP Chemistry exam with 100% brand new content that reflects recent exam changes Addressing the major overhaul that the College Board recently made to the AP Chemistry exam this AP Chemistry test prep guide includes completely brand new content tailored to the exam administered every May Features of the guide include review sections of the six big ideas that the new exam focuses on Fundamental building blocks Molecules and interactions Chemical reactions Reaction rates Thermodynamics Chemical equilibrium Every section includes review questions and answers Also included in the guide are two full length practice tests as well as a math review section and

sixteen discrete laboratory exercises to prepare AP Chemistry students for the required laboratory experiments section on the exam

Thermochemical Studies Semiannual Report Argonne National Laboratory. Chemical Engineering Division, 1972

Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16

Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program. The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure, selecting the proper apparatus, employing the proper techniques, systematically analyzing and interpreting the data, and minimizing inherent variables. As a result of good data, a scientific and analytical conclusion is made, which may or may not be right but is certainly consistent with the data.

Experiments write textbooks; textbooks don't write experiments. A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered. Further experimentation provides additional cause-effect observations leading to an even better understanding of the experiment. The 11th edition's experiments are informative and challenging while offering a solid foundation for technique, safety, and experimental procedure. The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment. The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit. An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment. Continued emphasis on handling data is supported by the Data Analysis section.

NASA Scientific and Technical Reports United States. National Aeronautics and Space Administration Scientific and Technical Information Division, 1965

Annual Report of the Director of the Geophysical Laboratory Carnegie Institution of Washington. Geophysical Laboratory, 1989

U.S. Government Research Reports, 1963

A Selected Listing of NASA Scientific and Technical Reports for ... United States. National Aeronautics and Space Administration. Scientific and Technical Information Division, 1964

Lewis Basicity and Affinity Scales Christian Laurence, Jean-François Gal, 2009-12-23

The Lewis concept of acids and bases is discussed in every general organic and inorganic chemistry textbook. This is usually just a descriptive treatment as it is not possible to devise a single numerical scale suitable for all occasions. However, quantitative Lewis acid-base chemistry can be developed by compiling reaction-specific basicity scales which can be used in specific branches of chemistry and biochemistry.

Lewis Basicity and Affinity Scales Data and Measurement brings together for the first time a comprehensive range of Lewis basicity/affinity data in one volume. More than 2400 equilibrium constants of acid-base reactions, 1500 complexation enthalpies, and nearly 2000 infrared and ultraviolet shifts upon complexation are gathered together in 25 thermodynamic and spectroscopic scales of basicity and/or affinity. For each scale, the definition, the method of measurement, an exhaustive database, and a critical discussion are given. All the data have been critically examined; some have been re-measured; literature gaps have been filled by original measurements; and each scale has been made homogeneous. This collection of data will enable

experimental chemists to better understand and predict the numerous chemical physical and biological properties that depend upon Lewis basicity Chemometricians will be able to apply their methods to the data matrices constructed from this book in order to identify the factors which influence basicity and basicity dependent properties In addition measured experimental basicities and affinities are essential to computational chemists for the validation calibration and establishment of reliable computational methods for quantifying and explaining intermolecular forces and the chemical bond Lewis Basicity and Affinity Scales Data and Measurement is an essential single source desktop reference for research scientists engineers and students in academia research institutes and industry in all areas of chemistry from fundamental to applied research The book is a noteworthy piece of work and represents a timely and vast accumulation of knowledge regarding Lewis bases that brings together accurate thermodynamic and spectroscopic data on typical reference Lewis acids As such it should serve as a useful and general guide to basicity J AM CHEM SOC 2011 133 642 *Chemical Engineering Division Summary Report*,1962-10 List of Bureau of Mines Publications and Articles ... with Subject and Author Index United States. Bureau of Mines,1987 *Basic Laboratory Principles in General Chemistry* Fitzgerald B. Bramwell,1990 List of Bureau of Mines Publications and Articles ... with Subject and Author Index United States. Bureau of Mines. Branch of Editorial Services,1987
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