

Quiz 4 – Potential Energy Diagrams

1. The addition of a catalyst to a reaction provides an alternative mechanism with

- A. Lower activation energy and lower reaction rate
- B. Lower activation energy and higher reaction rate**
- C. Higher activation energy and lower reaction rate
- D. Higher activation energy and higher reaction rate

2. Consider the following Reaction



The correct equation including the heat term is

- A. $\text{N}_{2(g)} + \text{O}_{2(g)} + 90 \text{ kJ} \longrightarrow 2\text{NO}_{(g)}$
- B. $\text{N}_{2(g)} + \text{O}_{2(g)} + 180 \text{ kJ} \longrightarrow 2\text{NO}_{(g)}$**
- C. $\text{N}_{2(g)} + \text{O}_{2(g)} \longrightarrow 2\text{NO}_{(g)} + 90\text{kJ}$
- D. $\text{N}_{2(g)} + \text{O}_{2(g)} \longrightarrow 2\text{NO}_{(g)} + 180\text{kJ}$

3. A forward reaction has an activation energy of 50 kJ and a ΔH of -100 kJ.
The PE diagram, which describes this reaction, is

A.

4. Consider the following potential energy diagram

The Activation energy for the forward reaction is

- A. 25 kJ
- B. 50 kJ
- C. 75 kJ**
- D. 125 kJ

5. Consider the following reaction



Multiple Choice Questions Energy Diagrams

Paul Foglino



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Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book **Problems and Problem Solving in Chemistry Education**

Georgios Tsaparis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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