

Q1. Which one of the following is **not** a correct statement about vitamin C, shown below?



- A** It is a cyclic ester.
- B** It can form a carboxylic acid on oxidation.
- C** It decolourises a solution of bromine in water.
- D** It is a planar molecule.

(Total 1 mark)

Q2. In which one of the following mixtures does a redox reaction occur?

- A** ethanal and Tollens' reagent
- B** ethanoyl chloride and ethanol
- C** ethanal and hydrogen cyanide
- D** ethanoic acid and sodium hydroxide

(Total 1 mark)

Q3. Propanoic acid reacts with methanol in the presence of a small amount of concentrated sulphuric acid. The empirical formula of the ester formed is

- A** CH₃O
- B** C₃H₇O₂
- C** C₃H₅O₂
- D** C₃H₇O

(Total 1 mark)

Multiple Choice On Carboxylic Acids

S Marginson



Multiple Choice On Carboxylic Acids:

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