PHYSICAL CHEMISTRY

(Complete Syllabus)

SECTION - A

Straight Objective Type

This section contains 220 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which ONLY ONE is correct.

Choose the correct answer:

The mass of oxygen required to produce enough CO which completely reduces x gm of Fe,O, Suppose molecular weight of O2 = y, molecular weight of $Fe_2O_3 = z$

$$(1) \quad \frac{2xa}{3y}$$

$$(2) \quad \frac{3\infty}{2z}$$

$$(3) \quad \frac{2y}{3xz}$$

$$(4) \quad \frac{3z}{2xy}$$

- The number of lone pair of electrons present in 2.56 am of orthorhombic sulphur will be
 - (1) 0.02 N_a

(2) 0.16 N_a

(3) 0.32 N_o

- (4) 0.08 N_a
- A 10.0 g of sample of mixture of calcium chloride and sodium chloride is treated with Na,CO, to precipitate the calcium as calcium carbonate. This CaCO, is heated to convert all calcium to CaO and final mass of CaO is 1.62 cms. The % by mass of CaCl, in the original mixture is
 - (1) 15.2%
 - (2) 32,1%
 - (3) 21.8%
 - (4) 11.07%
- 1000 ml of oxygen at STP were passed through an ozoniser and the resulting volume was 888 ml at STP. This quantity of ozonised oxygen is passed through excess of KI solution. The weight of I., liberated is
 - (1) 25.4 a

(2) 12.7 a

(3) 2.54 a

(4) 1.27 a

x, ml solution having normality 1 is diluted to get a solution of normality y,. What would be volume required for dilution?

(1)
$$x_1 \left(1 - \frac{y_1}{y_2}\right)$$
 (2) $x_1 \left(\frac{y_2}{y_1} - 1\right)$

(2)
$$x_1 \left[\frac{y_2}{y_1} - 1 \right]$$

(3)
$$x_1 \left(\frac{y_1}{y_2} - 1 \right)$$
 (4) $x_1 \left(1 + \frac{y_1}{y_2} \right)$

(4)
$$x_1 \left(1 + \frac{y_1}{y_2}\right)$$

- 1.625 g of an interhalogen compound ICI, was subjected to electric discharge which split it into constituents and brought to STP. It occupied 112 ml. The interhalogen compound is
 - (1) KCI

(2) ICL,

(3) ICL,

- (4) ICI,
- If 12.25 g of a sample of KCIO, of 75% purity and rest vapourisible impurties is strongly heated to give KCI, KCIO, and 0.32 gm O,. The mass of residue obtained is
 - (1) $1.769 \, \alpha$ (3) 5.327 g

- (2) 7.096 g (4) 8.85 g
- 8. A 2 g sample of xenon reacts with fluorine. The mass of compound produced is 3.158 g. The empirical formula of the compound is
 - (1) XeF.

(2) XeF. (4) XeF_e

- (3) XeF₆
- 1 g of carbonate of metal was dissolved in 25 ml
- of 1 N HCl. The resulting liquid required 5 ml of 1 N NaOH for neutralisation. The equivalent weight

Multiple Choice Physical Chemistry Exam Questions

R P Meena

Multiple Choice Physical Chemistry Exam Questions:

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The ordering of the chapters reflects the natural flow of events in an evolving Universe Philosophy of Science the basis of the view that natural events have natural causes Cosmology the origin of everything from the Big Bang to the current state of the Universe Geoscience the physics and chemistry behind the evolution of the planet Earth from its birth to the present Life Science the molecules and mechanisms of life on Earth Ecology the interdependence of all components within the Ecosphere and the Universe Information Content emphasis on how words and phrases and framing of issues affect opinions reliability of sources and the limitations of knowledge Addresses the four Ws of science Why scientists believe Nature works the way it does Who helped develop the fields of science What theories of natural processes tell us about the nature of Nature and Where our scientific knowledge is taking us into the future Gives a historical review of the evolution of science and the accompanying changes in the philosophy of how science views the nature of the Universe Explores the physics and chemistry of Nature with minimal reliance on mathematics Examines the structure and dynamics of the Universe and our Home Planet Earth Provides a detailed analysis of how humans as members of the Ecosphere have influenced and are continuing to influence the dynamics of events on the paludarium called Earth Presents underlying science of current political issues that shape the future of humankind Emphasizes how words and phrases and framing of issues can influence the opinions of members of society Makes extensive use of metaphors and everyday experiences to Physics, Chemistry & Biology Solved Papers (2023-24) illustrate principles in science and social interactions NEET/AIPMT) YCT Expert Team, 2023 24 NEET AIPMT Physics Chemistry Biology Solved Papers **NEET Exam** Chemistry Question Bank R P Meena, NEET Exam Chemistry Question Bank Topic wise and NCERT Based MCQs NEET Chemistry is considered to be the most scoring subject As it is one of the easiest scoring subjects it is often ignored and undermined a subject But if you want to get an edge over others here is a tip master NEET Chemistry concepts Chemistry demands the attention of students in understanding reactions and concrete basic understanding but once done it only gets easier from there NEET is amongst one the most prestigious medical entrance exams in India With just a few months left for the examination it becomes quite challenging for students to cover all the concepts included in the NEET syllabus thoroughly However a proper study plan designed as per the latest examination pattern and the syllabus can help students to prepare all the important concepts in shorter time duration Given below are few useful tips that can assist the students in tackling multiple choice questions in NEET exam accurately In most of the multiple choice questions the options are designed in a very tricky and confusing manner In most of the cases all the given options seem to be correct in some aspect Therefore the students are advised to read the entire question very carefully Try to accumulate all the information provided in the question effectively because in some of the cases you can easily evaluate the Answers from the question itself If you are muddled by the given options then give each option a true and false test Instead of getting confused consider all the possibilities and neglect the incorrect options Hence in this way the most appropriate answer could be easily spotted Topic wise MCQs for NEET Chemistry Environmental Chemistry Organic Chemistry P Block Elements States of Matter Redox Reaction Chemical Equilibrium Thermodynamics Periodic Classification of Elements s Block Elements Chemical Bonding Molecular Structure Structure of Atom Some Basic Concepts of Chemistry Hydrocarbons Some Basic Concepts of Chemistry Solid State Structure of Atom Solutions Classification of Elements and Periodicity 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the questions asked in the NEET exam are application based Therefore students need to focus more on the concepts along with its applications in order to score well in the examination Chemistry topics broadly can be grouped into Organic Inorganic and Physical chemistry Organic Chemistry is a vast section and holds the maximum weightage Its concepts and principles are important One needs to be conversant with properties of groups conversions recognizing organic compounds etc Interlinking between different chapters from organic chemistry can be of great help The students must primarily focus on reading NCERT textbooks Several times the questions asked in NEET exam are taken directly from the NCERT textbooks Initially avoid answering those questions for which you are not confident because your wrong answer may reduce your final score In order to utilize your time appropriately divide the three hours of examination time as per your comfort among Physics Chemistry and Biology Initially focus on attempting all easy questions and later on pick the difficult ones By this way your confidence will be elevated and you will also get more time to answer hard questions Practice previous

years question papers mock tests and sample papers to get an idea on how to answer MCQ questions efficiently Preparing at an early stage is what an MCQ exam requires Avoid guesswork for negative marking questions as they might lower your final score These tips can be very helpful for students to answer difficult and brain teaser questions Prior preparations and practice are mandatory aspects of any examination Hence to crack highly competitive examination like NEET it is mandatory for students to prepare well and acquire the skills to tackle multiple choice questions effectively Rather than just following mere guesswork the aspirants can focus on the tips discussed to tackle Multiple Choice Questions in NEET in the right 2024-25 NEET/AIPMT Physics Solved Papers YCT Expert Team , 2024 25 NEET AIPMT Physics Solved Papers Bilingual 592 895 This book contains 48 sets solved papers and 2275 objective questions Approach for Introductory Physical Chemistry Calculations Brian Murphy, Clair Murphy, Brian J. Hathaway, 1997 A Working Method Approach for Introductory Physical Chemistry Calculations is a concise inexpensive introduction to first year chemistry that is aimed at students who are weak in chemistry or have no chemistry on entry to university Such students usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty. The text should also be invaluable to first year intending chemists. This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical approach to problem solving in physical chemistry teaching them to think for themselves when faced with a problem Pearson Guide to Physical Chemistry for the IIT JEE Singhal, OCR A Level Year 2 Chemistry A Student Guide: Module 5 Mike Smith, 2016-05-16 Exam Board OCR Level A level Subject Chemistry First Teaching September 2015 First Exam Summer 2017 Written by experienced author Mike Smith this Student Guide for Chemistry Identifies the key content you need to know with a concise summary of topics examined in the A level specifications Enables you to measure your understanding with exam tips and knowledge check questions with answers at the end of the guide Helps you to improve your exam technique with sample answers to exam style questions Develops your independent learning skills with content Atkins' Physical Chemistry Peter William Atkins, Julio De Paula, James you can use for further study and research Keeler, 2023 Revised edition of Atkins Physical chemistry Peter Atkins Julio de Paula James Keeler Eleventh edition 2018 OCR A Level Year 2 Chemistry A Student Guide: Module 6 Mike Smith, 2016-06-06 Exam Board OCR Level A level Subject Chemistry First Teaching September 2015 First Exam June 2017 Written by experienced author Mike Smith this Student Guide for Chemistry Helps students identify what they need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to

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