

Determination of concentration of a Dextrose solution of unknown strength using an Ostwald Viscometer.

Aim: To determine the concentration of a Dextrose solution of unknown strength using an Ostwald Viscometer and to study the variation of this solution's viscosity as a function of temperature.

Apparatus Required: 1. Ostwald Viscometer, Thermometer  
2. Water, Ice water, glass jar  
3. Hot plate, standard volumetric flask 25 ml

Chemicals Required: 1. Dextrose, Distilled water

Theory: Viscosity: The internal friction of a liquid which resist the flow of it is known as viscosity.

The force it implies can be measured in terms of the tangent force which is needed to keep the speed of layers constant, this force is called viscous force.

$$\eta = \frac{\pi \mu^4 P L}{8 \pi V L}$$

$$\text{Volume rate law} = \frac{V}{t} = \frac{P \pi \mu^4}{8 \eta L}$$

$V$  = Volume of the liquid

$t$  = time of flow of liquid

$\eta$  = Viscosity coefficient

$P$  = Hydrostatic pressure

$L$  = Distance travelled by the liquid during time  $t$ .

The units of viscosity are called Poise (P). The official SI unit for absolute viscosity is kg/ms or (Pascal second, Pa.s).

Ostwald viscometer is a commonly used viscometer, which consists of a U-shaped glass tube held vertically. For more accurate

# Msc Organic Chemistry Practical Manual

**Kathleen Armour**



## **Msc Organic Chemistry Practical Manual:**

**ORGANIC MEDICINAL CHEMISTRY PRACTICAL MANUAL FOR PHARMACY AND SCIENCE STUDENTS** Anees Ahmad Siddiqui, The manual illustrates the concept of basic techniques in practical organic medicinal chemistry It aims to meet the requirements of B Pharmacy students under the new syllabus prescribed by Pharmacy Council of India It will also be useful to BSc BSc Hons and MSc medicinal chemistry students A Systematic Course of Practical Organic Chemistry Lionel Guy Radcliffe, 1905 **Laboratory Manual of Organic Chemistry** Raj K. Bansal, 2009 About the Book The manual has been thoroughly revised several new experiments and tests have been added while some redundant material has been deleted Chapter 2 has been completely rewritten An obvious change of this edition constitutes the splitting of Chapter 7 into two separate Chapters Tables on derivatives of organic compounds have been expended Also included are 20 estimations 75 preparations and isolation experiments and approximately 135 in text questions related to the experiments The approximation of modern spectroscopic techniques to structure determination have been discussed in the last Chapter This book is designed both for undergraduate and postgraduate level students with its enhanced and comprehensive presentation This is an indispensable book for organic chemistry practicals About the Author Dr Raj K Bansal received his M S from the University of California Davis Calif U S A and Ph D from Calgary University Calgary Alberta Canada He was a postdoctoral fellow at the National Research Council N R C of Canada in Halifax N S Canada followed by a Research Associateship at the Mellon Institute of Science Carnegie Mellon University Pittsburgh Pa U S A Dr Bansal has published a number of research papers in various foreign and Indian scientific journals He is the author of six books on chemistry including this work A Textbook of Organic Chemistry 5th ed 2007 Organic Chemistry Problems and Solutions 2nd edn 2006 and Heterocyclic Chemistry 4th edn 2005 One of his books Synthetic Approaches in Organic Chemistry has been reprinted by Jones and Bartlett Publishers Sudbury Massachusetts U S A Dr Bansal was a former Professor Department of Chemistry Indian Institute of Technology Delhi Hauz Khas New Delhi **Classified Guide to Technical and Commercial Books**, 1904 A Manual of Machine Drawing and Design David Allan Low, Alfred William Bevis, 1898 **Laboratory Manual in Engineering Chemistry : For the Students of JNTU Hyderabad** Dr. B. Rama Devi & Dr. P. Aparna, The book is written to gain the basic knowledge on the principles of chemistry required for practical applications in engineering concepts This book consists organic and general chemistry experiments for chemical engineering for 1st and 2nd semester students The book also explains the precautions and safety rules for avoiding the accidents in chemistry laboratory It covers Estimation of Ferrous iron by Dichrometry and Permanganometry Method Estimation of Acetic Acid by Conductometric Titrations Estimation of the Amount of Fe 2 by Potentiometry Determination of an Acid Concentration using pH Meter Preparation of Nylon 6 and Bakelite Phenol Formaldehyde Resin Estimation of Acid Value of Given Lubricant Oil Determination of Rate of Corrosion of Mild Steel Preparation of Benzanilide from Benzophenone via the Oxime by Beckmann Rearrangement etc **A Text-book**

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