

# MCQ on Periodicity of Elements

1. The modern periodic table based on - **Atomic number**
2. On the basis of atomic radii which one is correct? -  **$F < O < N$**
3. Group 18th elements is also known as - **Noble gases**
4. The most reactive element is - **Fluorine**
5. Transition metals belongs to - **Group number 3 to 12**
6. Which one has two completely filled shell? - **Neon**
7. Moving down the group of the periodic table, the factor that remains constant is - **Valence electrons**
8. Select the one that belongs to the same group as that of fluorine - **Astatine**
9. Which of the following is correct order of metallic character following elements is -  **$K > Mg > Al > B$**
10. The element with atomic number 33 is from - **Group 15**
11. In the long form of the periodic table, the transition metals belong to - **d-block**
12. The correct order of their non-metallic character is -  **$F > N > C > B > Si$**
13. The number of elements present in the 4th period - **18**
14. What would be the symbol for the element with atomic number 118? - **uuo**
15. Which of the following is not an actinoid? - **Terbium ( $Z = 65$ )**

## Chemistry Objective Questions

# Mcq With Explanation In Inorganic Chemistry

**Career Point Kota**



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Basic Principles of Inorganic Chemistry Brian Murphy, Clair Murphy, Brian J. Hathaway, 1998 General chemistry textbooks are usually lengthy and present chemistry to the student as an unconnected list of facts In inorganic chemistry emphasis should be placed on the connections between valence shell electron configuration and the physical and chemical properties of the element Basic Principles of Inorganic Chemistry Making the Connections is a short concise book that emphasises these connections in particular the chemistry of the Main Group compounds With reference to chemical properties Lewis Structures stoichiometry and spider diagrams students will be able to predict or calculate the chemistry of simple polyatomic compounds from the valence shell configuration and will no longer be required to memorise vast amounts of factual chemistry This book is ideal for students taking chemistry as a subsidiary subject as well as honours degree students

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