

Wheelchairs

McKesson wheelchairs are ideal for patients who have difficulty walking due to illness, injury or disability. McKesson offers a comprehensive selection of wheelchairs that promote patient independence, improve mobility and enhance quality of life. With durable construction, easy-to-adjust footrests and lightweight, easy-to-clean upholstery, McKesson wheelchairs provide long-lasting performance and comfort. Dual axle options provide easy transitions to hemi-level seat height to allow users to propel the wheelchair with their feet. Choose a McKesson wheelchair based on arm style, footrest preferences, reclining options, size and weight capacity.

Features and Benefits

Standard Wheelchairs

- Carbon steel frame with a textured, silver-vein finish for enhanced visual appeal and ease of maintenance
- Embossed vinyl upholstery is durable, lightweight, attractive and easy-to-clean
- Urethane tires are mounted on composite wheels to provide durability and a smooth ride over most surfaces
- 8-inch front casters are adjustable in three positions
- Convenient carry pouch for transporting personal items
- Not made with natural rubber latex



Lightweight Wheelchairs

- Carbon steel frame with a textured, silver-vein finish for enhanced visual appeal and ease of maintenance
- Nylon upholstery is lightweight, and breathable for user's lasting comfort
- Urethane tires are mounted on mag-style, composite wheels to provide durability and a smooth ride over most surfaces
- Precision-sealed wheel bearings in front and rear ensure long-lasting performance and reliability
- 8-inch front caster wheels provide a smooth ride over most surfaces
- Flipback attachable armrests for patient comfort and easy transfer
- Not made with natural rubber latex



Satisfaction Guaranteed

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Mckesson Paragon User Manual

ML Morrison



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web using algebra the moles was founded to be 0.00352 moles of butane in order to find the molarity the mass of the butane should be divided by the moles after further calculations the molarity of the butane gas was calculated to be 56.81 grams moles this lab is done with tons of sources of error

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web this data shows that there was a gain the mass of the lighter which shows that there was a certain amount of butane 0.38 grams 0.02 grams gas that was successfully collected during the gas collection there is a mild influence of barometric pressure during both trials as shown above in the raw data

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web 11 use avogadro's law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22.4 l at STP $V_1/n_1 = V_2/n_2$ $0.0896 \text{ l} / n_2 = 22.4 \text{ l} / 1 \text{ mole}$ $n_2 = 0.0040 \text{ moles}$ 12 experimental molar mass g/mole of butane 0.23 g / 0.0040 moles 58 g/mole 13 the molecular formula of butane is C_4H_{10}

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web nov 3 2020 in this experiment the moles and mass will be required to determine the molar mass of butane in a butane lighter experimental to begin the experiment the mass of a butane lighter was measured before using the butane gas to deplete the water to the 80ml mark on a graduated cylinder

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