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$\text{HNO}_3 + \text{Mg}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{H}_2\text{C}_2\text{H}_3\text{O}_2 + \text{Mg}(\text{NO}_3)_2$ chemical equation - Aug 11 2023

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web magnesium is already balanced on the reactant and product side balancing nitrogen by multiplying hno_3 by 2 we get $\text{mg} + 2\text{hno}_3 \rightarrow \text{mg}(\text{no}_3)_2 + \text{h}_2$ mg s magnesium 2 hno_3 1 nitric acid $\text{mg}(\text{no}_3)_2$ s magnesium nitrate h_2 g hydrogen gas oxygen and hydrogen get automatically balanced by balancing nitrogen

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