[4225] - 102

M.Sc.

INORGANIC CHEMISTRY CH - 130 : Inorganic Chemistry - I

(2008 Pattern) (Semester - I) (Part - I)

Time: 3 Hours! Instructions to the candidates: IMax. Marks: 80

- 11 All questions are compulsory.
- Answers to the two sections should be written in separate answer books. 2)
- Neat and labelled diagrams must be drawn whenever necessary. 3)
- 4) Figures to the right indicate full marks.

SECTION - I

Q1) Attempt any three of the following:

- [15]
- Sketch and explain the following symmetry elements in [B F.] ion and classify it into appropriate point group,
 - Plane of symmetry.

- Centre of inversion. m)
- b) Using great orthogonality theorem, derive the character table for D, point group. Given the classes of D, point group are, E, C2, C2, and C2.
- Develop stereographic projections for the following molecules. Justify your answer.
 - Staggered ferrocene.

- Cyclopentadienyl anion. 111)
- d) By schematic representations, give the products of following symmetry operations in NF, molecule.
 - i) $C_3 \times C_3^{-1}$. ii) $C_3 \times \sigma V_1$. iii) $\sigma V_1 \times \sigma V_2$.

Attempt any three of the following: (O2)

- [115]
- a) For a X_sF_s molecule, find out reducible representation for which sigma bonds form the basis and find out which of the orbitals from the X_ atom will be offered for sigma bonding.
- Given: Character table for D,h point group.
- Classify any two of the following molecules into appropriate point group. Justify it.
 - i) CO₂²⁻ (Planar).
- S.F.

iiii) SO.

Msc Chemistry Entrance Exam Previous Year Papers

William Shurtleff,H.T. Huang,Akiko Aoyagi

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