

Acids and Bases

How do acids and bases behave in water?

Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

Model 1 – Arrhenius Acids and Bases

| Common Name for Aqueous Solution | Chemical Formula | Found in... | Tastes... | Turns Litmus Paper... | Conducts Electricity? | Acid or Base |
|----------------------------------|---|-------------------|-----------|-----------------------|-----------------------|--------------|
| Acetic acid | $\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$ | Vinegar | Sour | Red | Yes | Acid |
| Benzoic acid | $\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$ | Food preservative | Sour | Red | Yes | Acid |
| Phosphoric acid | $\text{H}_3\text{PO}_4(\text{aq})$ | Soda pop | Sour | Red | Yes | Acid |
| Hydrochloric acid | $\text{HCl}(\text{aq})$ | Stomach acid | Sour | Red | Yes | Acid |
| Citric acid | $\text{H}_2\text{C}_6\text{H}_7\text{O}_7(\text{aq})$ | Citrus fruits | Sour | Red | Yes | Acid |
| Ascorbic acid | $\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$ | Vitamin C | Sour | Red | Yes | Acid |
| Magnesium hydroxide | $\text{Mg}(\text{OH})_2(\text{aq})$ | Milk of magnesia | Bitter | Blue | Yes | Base |
| Aluminum hydroxide | $\text{Al}(\text{OH})_3(\text{aq})$ | Antacids | Bitter | Blue | Yes | Base |
| Barium hydroxide | $\text{Ba}(\text{OH})_2(\text{aq})$ | Lubricants | POISON | Blue | Yes | Base |
| Sodium hydroxide | $\text{NaOH}(\text{aq})$ | Drain cleaner | POISON | Blue | Yes | Base |

- Refer to Model 1.
 - What is the common chemical name for Vitamin C?
ascorbic acid
 - Is Vitamin C classified as an acid or a base?
acid
- Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.
sour, red, conduct electricity
- Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
- Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

Introduction To Acids And Bases Pogil Key

ML Yell



Introduction To Acids And Bases Pogil Key:

The book delves into Introduction To Acids And Bases Pogil Key. Introduction To Acids And Bases Pogil Key is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Introduction To Acids And Bases Pogil Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Introduction To Acids And Bases Pogil Key
- Chapter 2: Essential Elements of Introduction To Acids And Bases Pogil Key
- Chapter 3: Introduction To Acids And Bases Pogil Key in Everyday Life
- Chapter 4: Introduction To Acids And Bases Pogil Key in Specific Contexts
- Chapter 5: Conclusion

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3. In chapter 2, this book will delve into the foundational concepts of Introduction To Acids And Bases Pogil Key. The second chapter will elucidate the essential principles that need to be understood to grasp Introduction To Acids And Bases Pogil Key in its entirety.
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5. In chapter 4, the author will scrutinize the relevance of Introduction To Acids And Bases Pogil Key in specific contexts. This chapter will explore how Introduction To Acids And Bases Pogil Key is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Introduction To Acids And Bases Pogil Key. The final chapter will summarize the key points that have been discussed throughout the book.
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