

Book 3: Example questions. All questions were translated from German. Keys are marked with an asterisk [*]. See text for details.

Question 1. Example of a kypnose-type question.

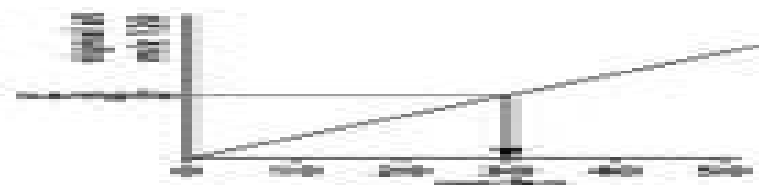
What increases the spectral resolution of an ICP-MS instrument?

- | | | |
|-----------------------|-----------------------|---|
| True | False | |
| ☐ | ☐ | Scatter plate in a sector field instrument |
| ☐ | ☐ | Using a dual mode detector with analog and pulse counting |
| ☐ | ☐ | Detection of barium at m/z 138 instead of 137 |
| ☐ | ☐ | Increase the length of the quadrupole for an ICP-QMS |

Question 2. Silver has a mass of 107.87 g/mol. At which m/z do you detect silver with ICP-MS?

- | | |
|----|-----|
| A | 106 |
| B* | 107 |
| C | 108 |
| D* | 109 |

Question 3. You quantify tin in a sample with ICP-MS and external calibration at m/z 120. The isotope abundance of ^{120}Sn is about 33%. The calibration for the signal at m/z 120 indicates a mass fraction of 30 $\mu\text{g}/\text{kg}$ [see figure below]. What is the mass fraction of tin in the measurement aliquot?



- | | |
|----|----------------------------|
| A | 90 $\mu\text{g}/\text{kg}$ |
| B | 60 $\mu\text{g}/\text{kg}$ |
| C* | 30 $\mu\text{g}/\text{kg}$ |
| D | 10 $\mu\text{g}/\text{kg}$ |

Question 4. Chocolate was digested in a closed vial under pressure using nitric acid and hydrogen peroxide using a microwave system. What constitutes the main matrix of the analyte cadmium in the prepared solution after the digestion of chocolate?

- | | |
|----|-------------------|
| A | cadm., sugar, fat |
| B* | water |
| C | cadmium ions |
| D | no matrix |

Question 5. How do you prepare a sample of selenium to differentiate between and quantify organic mercury species?

- | | |
|----|---|
| A | microwave digestion using HCl and H_2O_2 |
| B | microwave digestion using HNO_3 and H_2O_2 |
| C | microwave digestion using HNO_3 |
| D* | no digestion |

Multiple Choice Question For Instrumental Chemistry

David Chitate



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Table of Contents Multiple Choice Question For Instrumental Chemistry

1. Understanding the eBook Multiple Choice Question For Instrumental Chemistry
 - The Rise of Digital Reading Multiple Choice Question For Instrumental Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice Question For Instrumental Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Multiple Choice Question For Instrumental Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice Question For Instrumental Chemistry
 - Personalized Recommendations
 - Multiple Choice Question For Instrumental Chemistry User Reviews and Ratings
 - Multiple Choice Question For Instrumental Chemistry and Bestseller Lists
5. Accessing Multiple Choice Question For Instrumental Chemistry Free and Paid eBooks
 - Multiple Choice Question For Instrumental Chemistry Public Domain eBooks
 - Multiple Choice Question For Instrumental Chemistry eBook Subscription Services
 - Multiple Choice Question For Instrumental Chemistry Budget-Friendly Options
6. Navigating Multiple Choice Question For Instrumental Chemistry eBook Formats

- ePub, PDF, MOBI, and More
- Multiple Choice Question For Instrumental Chemistry Compatibility with Devices
- Multiple Choice Question For Instrumental Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Multiple Choice Question For Instrumental Chemistry
 - Highlighting and Note-Taking Multiple Choice Question For Instrumental Chemistry
 - Interactive Elements Multiple Choice Question For Instrumental Chemistry
- 8. Staying Engaged with Multiple Choice Question For Instrumental Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Multiple Choice Question For Instrumental Chemistry
- 9. Balancing eBooks and Physical Books Multiple Choice Question For Instrumental Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Multiple Choice Question For Instrumental Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Multiple Choice Question For Instrumental Chemistry
 - Setting Reading Goals Multiple Choice Question For Instrumental Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Multiple Choice Question For Instrumental Chemistry
 - Fact-Checking eBook Content of Multiple Choice Question For Instrumental Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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