



Physical Science

PDF Study Guide

Properties of Elements
 Study Guide: The Periodic Table, Ions, and...
 Questions and Answers
 1. Q: What are the main characteristics of metals?
 A: High electrical and thermal conductivity, and a shiny appearance.
 2. Q: Where are non-metals located on the periodic table?
 A: On the right side of the periodic table.
 3. Q: What is a noble gas?
 A: A group of gases in the periodic table.
 4. Q: Name a characteristic of noble gases.
 A: They are inert and rarely form compounds.
 5. Q: What type of elements are found in the middle of the periodic table?
 A: Metals.
 6. Q: What type of elements are found in the bottom of the periodic table?
 A: Halogens or noble gases.

Properties of Elements
 Study Guide: The Periodic Table, Ions, and Valence Electrons
 1. Overview of the Periodic Table
 The periodic table is a systematic arrangement of elements based on their atomic number, electron configurations, and recurring chemical properties. Elements are organized into vertical columns (groups) and horizontal rows (periods).
 2. Metals, Non-Metals, and Metalloids
 Metals are located on the left side of the periodic table. They are characterized by high electrical and thermal conductivity, malleability, and a shiny appearance. Examples include iron, copper, and aluminum.
 Non-metals are located on the right side of the periodic table. They are characterized by low electrical and thermal conductivity, brittleness, and a variety of physical states. Examples include oxygen, carbon, and sulfur.
 Metalloids are located between metals and non-metals. They have properties intermediate between metals and non-metals. Examples include silicon, boron, and arsenic.
 3. Noble Gases (Group 18)
 Noble gases are located on the far right of the periodic table. They are characterized by their extremely low reactivity, which is due to their full outer shell of electrons. Examples include helium, neon, argon, krypton, xenon, and radon.

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