

- Q.20 For a reversible reaction forward and backward rate constant respectively are 2.5×10^{-4} and 7.5×10^{-4} . What is the value of equilibrium constant [RPMT-2001]
 (1) 3 (2) $1/3$ (3) 1 (4) None
- Q.21 If the equilibrium constant for the reaction $N_2 + 3H_2 \rightarrow 2NH_3$ is K then the equilibrium constant for the reaction $2N_2 + 6H_2 \rightarrow 4NH_3$ would be equal to [RPMT-2002]
 (1) K^2 (2) \sqrt{K} (3) $\frac{1}{\sqrt{K}}$ (4) $\frac{1}{K^2}$
- Q.22 The reaction quotient (Q) for the reaction $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$ is given by $Q = \frac{[NH_3]^2}{[N_2][H_2]^3}$. The reaction will proceed from right to left if [CBSE-PMT-2003]
 (1) $Q < K_e$ (2) $Q > K_e$ (3) $Q = 0$ (4) $Q = K_e$
- Q.23 The following equilibria are given [CBSE-PMT-2003]
 $N_2 + 3H_2 \rightarrow 2NH_3$ K_1
 $N_2 + O_2 \rightarrow 2NO$ K_2
 $H_2 + \frac{1}{2}O_2 \rightarrow H_2O$ K_3
 The equilibrium constant of the reaction $2NH_3 + \frac{5}{2}O_2 \rightarrow 2NO + 3H_2O$, in terms of K_1 , K_2 and K_3 is :
 (1) $K_1 K_2 / K_3$ (2) $K_1 K_3^3 / K_2$ (3) $K_2 K_3^3 / K_1$ (4) $K_1 K_2 K_3$
- Q.24 For the gas phase reaction, $C_2H_4 + H_2 \rightleftharpoons C_2H_6$ ($\Delta H = -32.7$ kcal), carried out in a vessel, the equilibrium conc. of C_2H_4 can be increased by [AMU 2003]
 (1) Decreasing the pressure (2) Increasing the temperature
 (3) Removing some C_2H_6 (4) Adding some H_2
- Q.25 Consider the following reversible reaction at equilibrium, $2H_2O(g) \rightleftharpoons 2H_2(g) + O_2(g)$ $\Delta H = 24.7$ kJ. Which one of the following changes in conditions will lead to maximum decomposition of $H_2O(g)$? [Kerala CEE 2004]
 (1) Increasing both temperature and pressure
 (2) Decreasing temperature and increasing pressure
 (3) Increasing temperature and decreasing pressure
 (4) Increasing temperature at constant pressure
- Q.26 In the reaction, $A_2(g) + 4B_2(g) \rightleftharpoons 2AB_4(g)$ $\Delta H < 0$, the formation of $AB_4(g)$ will be favoured at [MP PMT 2004]
 (1) High temperature, low pressure (2) Low temperature, high pressure
 (3) High temperature, high pressure (4) Low temperature, low pressure

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web as a general rule the life time of the timing belt oscillates between 37280 mi and 99420 mi or 5 years whichever comes first check the owner's manual of your pacifica to know the exact data of your car when changing the belt replace the tensioners as well

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web the timing belt allows the crankshaft to drive the camshafts in the cylinder head at half the rpms of the crankshaft the camshafts then open and close the engine's intake and exhaust valves in time with the movement and position of the pistons in the engine when to replace the timing belt every 60k 90k miles

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web apr 21 2021 it appears to be the diagram for the older 2004 2006 pacifica suv with the 3.5l the dead giveaways are the power steering pump and the lack of water pump which is driven by the timing belt on the 3.5l i've attached the correct diagram below thanks for correcting that

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