

1st Semester Chemistry Final Exam Review

1. Identify the following reaction type AND correct products:



- a. combustion
- b. synthesis
- c. decomposition
- d. single replacement
- e. double replacement

- w. $\text{Ag} + \text{ScOH}$
- x. $\text{Ag} + 2 \text{Sc}(\text{OH})_2$
- y. $3 \text{Ag} + \text{Sc}(\text{OH})_3$
- z. $3 \text{Ag} + 2 \text{Sc}(\text{OH})_3$

- | | |
|-------|---------|
| I. | a. & w. |
| II. | b. & z. |
| III. | b. & y. |
| IV. | d. & z. |
| V. | e. & z. |
| VI. | d. & y. |
| VII. | e. & y. |
| VIII. | c. & z. |
| IX. | d. & x. |

2. Convert: $2.32 \times 10^5 \text{ mg} = ? \text{ kg}$

3. Calculate the energy for a quantum of light with a frequency of $8.44 \times 10^{18} \text{ Hz}$.

4. Determine the oxidation number for iodine in aluminum iodate.

5. REPORTING WITH THE CORRECT NUMBER OF SIGNIFICANT DIGITS, how many grams are in **4.011 moles** of calcium chloride?

- a) 445.14078 grams b) 445.140 grams c) 445.1 grams d) 445 grams
e) 450 grams f) 400 grams

6. Calculate the frequency of light that has a wavelength of 210 nm.

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Jessica J Manson



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 - The Rise of Digital Reading Mcps Honors Chemistry Review 2015 Bcrs Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Mcps Honors Chemistry Review 2015 Bcrs Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcps Honors Chemistry Review 2015 Bcrs Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mcps Honors Chemistry Review 2015 Bcrs Answers
 - Personalized Recommendations
 - Mcps Honors Chemistry Review 2015 Bcrs Answers User Reviews and Ratings
 - Mcps Honors Chemistry Review 2015 Bcrs Answers and Bestseller Lists
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 - Mcps Honors Chemistry Review 2015 Bcrs Answers Public Domain eBooks
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 - Highlighting and Note-Taking Mcps Honors Chemistry Review 2015 Bcrs Answers
 - Interactive Elements Mcps Honors Chemistry Review 2015 Bcrs Answers
8. Staying Engaged with Mcps Honors Chemistry Review 2015 Bcrs Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcps Honors Chemistry Review 2015 Bcrs Answers
9. Balancing eBooks and Physical Books Mcps Honors Chemistry Review 2015 Bcrs Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcps Honors Chemistry Review 2015 Bcrs Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mcps Honors Chemistry Review 2015 Bcrs Answers
- Setting Reading Goals Mcps Honors Chemistry Review 2015 Bcrs Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mcps Honors Chemistry Review 2015 Bcrs Answers
- Fact-Checking eBook Content of Mcps Honors Chemistry Review 2015 Bcrs Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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