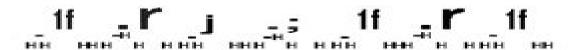
- 3.50 To solve this type of problem, read the problem carefully, word for word. Then by to interpret parts of the problem. For example:
 - 1) Formaldebyde is so aldebyde, H-C-H
 - It trimerizes that is, 3 formuldebydes come together to form a compound C5H603.
 Because no atoms are eliminated, 68 of the original atoms are still present.
 - 3) There are no carbonyl groups. This means that trioune cannot contain any -C=O functional groups. If you look back to Table 3.1, you can see that the only oxygen-containing functional groups that can be present are either others or alcohols.
 - 4) A monohemmo derivative is a compound in which one of the -H's has been replaced by a -Br. Became only one monohemno derivative is possible, we know that there can only be one type of bydrogen in trioxime. The only possibility for trioxine is:

3.51 The highest energy conformation of bosonorthane has a strain energy of 15 kJ/mol. Because this includes two H-H eclipsing interactions of 4.0 kJ/mol each, the value of so H-Br sclipsing interaction is 15 kJ/mol. 2(4.0 kJ/mol) = 7.0 kJ/mol.

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