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# Chemistry 12 Worksheet 4-2

82

# Bronsted Acids and Equilibria

and the same of th	
21	Write the formula for a proton (1 mark)
2.	Write the formula for a hydrated proton (I mark) H30 <sup>†</sup>
3.	Write the formula for a hydronium ion(1 mark) H30 <sup>+</sup>
4.	Give the Arrhenius definition of an acid(1 mark) any subst- which
	releases H+10ns in water
5.	Veleases OH 1005 in water.
0	donates protons to another species which
74	Give the Bronsted definition of a base (1 mark) a Species which
0	accepts protons from another species
8.	Given the equation: HCO <sub>2</sub> + H <sub>2</sub> S ≠ H <sub>2</sub> CO <sub>3</sub> + HS
	a) The acid on the left side is (1 mark) H <sub>2</sub> S
_	b) The base on the left side is (1 mark) HCO3
11/	c) The acid on the right side is (1 mark) H <sub>2</sub> CO <sub>3</sub>
1.1	d) The base on the right side is (1 mark) HS

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**RD Boyd** 

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Improving High School Students' Performance in Chemistry with a Hands-on Approach Mary L. Fredell,1998
Introduction to Chemistry, Laboratory Manual T. R. Dickson,1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

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