

Acids and Bases

Question 1: Choose the correct answer for each of the following questions:

- 1.1 The element present in all acids...
- | | |
|------------|------------------|
| A oxygen | B carbon dioxide |
| C nitrogen | D hydrogen |
- 1.2 A sample of an acid in a beaker has a pH of 4. An alkali is added to it. What is the pH of the liquid in the beaker?
- | | |
|-------------|------------------|
| A decreases | B stays the same |
| C increases | D neutral |
- 1.3 Choose the incorrect combination:
- | | |
|-------------------------------------|---|
| A HNO_3 – Nitric acid. | B H_2SO_4 – sulphuric acid. |
| C HCl – Hydrochloric acid. | D H_2CO_3 – carbonic acid. |
- 1.4 The colour of blue litmus paper in an acid is ...
- | | |
|----------|---------|
| A yellow | B green |
| C blue | D red |
- 1.5 Which substance is always produced in the reaction between hydrochloric acid and lithium?
- | | |
|--------------|----------------------|
| A water | B hydrogen gas |
| C oxygen gas | D carbon dioxide gas |
- 1.6 Which substance is always produced in the reaction between hydrochloric acid and sodium hydroxide?
- | | |
|--------------|----------------|
| A water | B hydrogen gas |
| C oxygen gas | D precipitate |
- 1.7 Which formula represents an acid?
- | | |
|--------------------------|----------------------------|
| A KOH | B KCl |
| C CH_3OH | D CH_3COOH |
- 1.8 The method of adding an acid to a base is:
- | | |
|------------------|-----------------|
| A neutralisation | B precipitation |
| C evaporation | D decomposition |

[8]

Question 2: Write the general reaction for an acid with the following compounds:

- 2.1 Base: _____
- 2.2 Carbonate: _____
- 2.3 Metal: _____

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Mark Steyvers



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Table of Contents Grade 9 Natural Science Final Test 2014

1. Understanding the eBook Grade 9 Natural Science Final Test 2014
 - The Rise of Digital Reading Grade 9 Natural Science Final Test 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 9 Natural Science Final Test 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 9 Natural Science Final Test 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 9 Natural Science Final Test 2014
 - Personalized Recommendations
 - Grade 9 Natural Science Final Test 2014 User Reviews and Ratings
 - Grade 9 Natural Science Final Test 2014 and Bestseller Lists
5. Accessing Grade 9 Natural Science Final Test 2014 Free and Paid eBooks

- Grade 9 Natural Science Final Test 2014 Public Domain eBooks
- Grade 9 Natural Science Final Test 2014 eBook Subscription Services
- Grade 9 Natural Science Final Test 2014 Budget-Friendly Options
- 6. Navigating Grade 9 Natural Science Final Test 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 9 Natural Science Final Test 2014 Compatibility with Devices
 - Grade 9 Natural Science Final Test 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 9 Natural Science Final Test 2014
 - Highlighting and Note-Taking Grade 9 Natural Science Final Test 2014
 - Interactive Elements Grade 9 Natural Science Final Test 2014
- 8. Staying Engaged with Grade 9 Natural Science Final Test 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 9 Natural Science Final Test 2014
- 9. Balancing eBooks and Physical Books Grade 9 Natural Science Final Test 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 9 Natural Science Final Test 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 9 Natural Science Final Test 2014
 - Setting Reading Goals Grade 9 Natural Science Final Test 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 9 Natural Science Final Test 2014
 - Fact-Checking eBook Content of Grade 9 Natural Science Final Test 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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