Naming Ionic Compounds - Answer Key

Give the name of the following ionic compounds:

Name

| 1) | Na ₂ CO ₃ | sodium carbonate |
|-----|-----------------------------------|-----------------------------------|
| 2) | NaOH | sodium hydroxide |
| 3) | MgBr ₂ | magnesium bromide |
| 4) | KCI | potassium chloride |
| 5) | FeCl ₂ | iron (II) chloride |
| 6) | FeCl ₃ | iron (III) chloride |
| 7) | Zn(OH) ₂ | zinc hydroxide |
| 8) | Be ₂ SO ₄ | beryllium sulfate |
| 9) | CrF ₂ | chromium (II) fluoride |
| 10) | AJ_2S_3 | aluminum sulfide |
| 11) | PbO | lead (II) oxide |
| 12) | Li ₃ PO ₄ | lithium phosphate |
| 13) | Titla | titanium (IV) iodide |
| 14) | Co ₃ N ₂ | cobalt (II) nitride |
| 15) | Mg ₃ P ₂ | magnesium phosphide |
| 16) | Ga(NO ₂) ₃ | gallium nitrite |
| 17) | Ag ₂ SO ₃ | silver sulfite |
| 18) | NH ₄ OH | ammonium hydroxide |
| 19) | AI(CN) ₃ | aluminum cyanide |
| 20) | Be(CH ₃ CO | O) ₂ beryllium acetate |
| | | |

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Dennis Arden Hooker

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