

Question	Answer	Mark	Guidance
1 (a)	(The enthalpy change that accompanies) the formation of one mole of (an ionic) compound ✓ from its gaseous ions ✓ (under standard conditions)	2	IGNORE 'Energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole 2nd mark requires gaseous ions IF candidate response has '1 mole of gaseous ions', award 2nd mark but NOT 1st mark IGNORE: $\text{Mg}^{2+}(\text{g}) + 2\text{Cl}^{-}(\text{g}) \longrightarrow \text{MgCl}_2(\text{s})$ (question asks for words)
(b) (i)	Hydration involves bond forming OR bonds are made ✓	1	ALLOW statement of any type of bond being formed ALLOW (chloride) ions attract water (molecules) ALLOW a response in terms of hydrogen bonds breaking AND bond making DO NOT ALLOW response stating that energy is required DO NOT ALLOW response that refers to ions in H_2O , eg H^+
(ii)		2	Correct species AND state symbols required for both marks Mark each marking point independently ALLOW response on upper line: $\text{Mg}^{2+}(\text{g}) + 2\text{Cl}^{-}(\text{aq})$ (ie Cl^{-} hydrated before Mg^{2+}) ALLOW $\text{MgCl}_2(\text{aq})$

F325 June 13 Past Paper

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