

Aromatic Substitution Reactions

1. The correct systematic name for the aromatic compound shown is



(b) 1-ethyl-3-methylbenzene is substituted

2. Which is different? Substituents showing more resonance suggest the lowest

- (a) benzene substituted by the electron-withdrawing substituents giving higher π is often electron rich with
- (b) groups with the benzene ring is in a ring with electron groups
- (c) there is a lone electron substituent (lone) which is not a group



3. Which is different? Substituents showing more resonance suggest the lowest

4. Which is different? Substituents showing more resonance suggest the lowest

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- (b) groups with the benzene ring is in a ring with electron groups
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5. Which is the correct systematic name for the aromatic compound shown?



(b) 1,3-dimethylbenzene

6. Which is the correct systematic name for the aromatic compound shown?



- (b) 1,3-dimethylbenzene
- (c) 1,3-dimethylbenzene
- (d) 1,3-dimethylbenzene

7. Which is the correct systematic name for the aromatic compound shown?

- (a) 1,3-dimethylbenzene
- (b) 1,3-dimethylbenzene
- (c) 1,3-dimethylbenzene
- (d) 1,3-dimethylbenzene

8. Which is the correct systematic name for the aromatic compound shown?

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LP Steffe



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