

SECOND YEAR HIGHER SECONDARY MODEL EXAMINATION MARCH 2022

SUBJECT: CHEMISTRY

Qn. Code: ME 525

Qn. No.	Sub Qns.	Answer Key/Value Points	Score	Total												
PART I																
A. Answer any 5 questions from 1 to 9. Each carries 1 score																
1.		(b) Co	1	1												
2.		38% Sulphuric acid (H ₂ SO ₄) solution	1	1												
3.		(d) Molarity	1	1												
4.		s ⁻¹	1	1												
5.		Nickel (Ni)	1	1												
6.		(c) Rubber Latex	1	1												
7.		(a) CH ₃ -NH ₂	1	1												
8.		CH ₃ -CH ₂ -OH (Ethanol)	1	1												
9.		(c) COCl ₂	1	1												
B. Answer all questions from 10 to 13. Each carries 1 score																
10.		(b) Thymine	1	1												
11.		(b) Phenol, formaldehyde	1	1												
12.		(c) Artificial Sweetener	1	1												
13.		(b) Zinc	1	1												
PART II																
A. Answer any 2 questions from 14 to 17. Each carries 2 scores																
14.		<table><tr><th></th><th>Order</th><th>Molecularity</th></tr><tr><td>1.</td><td>It is the sum of the powers of the concentration terms in the rate law expression</td><td>It is the total number of reactant species collide simultaneously in a chemical reaction</td></tr><tr><td>2.</td><td>It is an experimental quantity</td><td>It is a theoretical quantity</td></tr><tr><td>3.</td><td>It can be zero or fractional</td><td>It cannot be zero or fractional</td></tr></table>		Order	Molecularity	1.	It is the sum of the powers of the concentration terms in the rate law expression	It is the total number of reactant species collide simultaneously in a chemical reaction	2.	It is an experimental quantity	It is a theoretical quantity	3.	It can be zero or fractional	It cannot be zero or fractional	2	2
			Order	Molecularity												
		1.	It is the sum of the powers of the concentration terms in the rate law expression	It is the total number of reactant species collide simultaneously in a chemical reaction												
		2.	It is an experimental quantity	It is a theoretical quantity												
3.	It can be zero or fractional	It cannot be zero or fractional														
(Any 2 differences required)																
15.		The regular decrease in the atomic and ionic radii along lanthanide series is known as lanthanide contraction.	1	2												
		Consequences: i) Due to Lanthanide Contraction the 2nd and 3rd row transition series elements have similar radii. ii) Lanthanides have similar physical properties and they occur together in nature. So their isolation is difficult. [Any one required]	1													
16.		Hinsberg reagent is benzene sulphonyl chloride (C ₆ H ₅ SO ₂ Cl). It is used to distinguish the three types of amines.	1 1	2												
17.		Osmotic pressure (π) = CRT Here C = 0.1 M, R = 0.082 Latm/K/mol and T = 27°C = 27 + 273 = 300 K So, π = 0.1 × 0.082 × 300 = 2.46 atm	1 1	2												
B. Answer any 2 questions from 18 to 20. Each carries 2 scores																
18.		In conductors, the valence band is either partially filled or it is overlapped with the conduction band. So electrons can easily flow from valence band to conduction band. In insulators, there is a large gap between valence band and conduction band. So no	2	2												

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