

Chapter 22

Hydrocarbon Compounds

22.1 Hydrocarbons

Organic Chemistry and Hydrocarbons

- Some thought that only living organisms could synthesize the organic compounds found in living cells.
- Synthetic chemical products (Plastics, dyes, etc.) are made from organic substances via synthesis, creating organic compounds that do not exist.
- Today, organic chemistry involves the discovery of almost all carbon compounds, regardless of origin.

Introduction to Hydrocarbons

Hydrocarbon compounds are called hydrocarbons.

Hydrocarbon is an organic compound composed of carbon and hydrogen.

Hydrocarbons have two carbon skeletons, as carbon atoms always form four covalent bonds.

Hydrocarbons are organic compounds composed of carbon and hydrogen atoms, forming the basis of many natural products.

Hydrocarbons are the simplest of all organic compounds. They are the basis of many natural products, such as fuels, plastics, and medicines.

They are the simplest of all organic compounds. They are the basis of many natural products, such as fuels, plastics, and medicines.



The Properties of Hydrocarbons

Hydrocarbons are organic compounds.

Hydrocarbons are organic compounds that contain only carbon and hydrogen atoms. They are the simplest of all organic compounds.

Hydrocarbons and the Environment

Hydrocarbons are organic compounds that contain only carbon and hydrogen atoms. They are the simplest of all organic compounds.

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