

FOE YENCE

Universal Compact Digital
Pressure Sensor
AP-030 (P) Series

Instruction Manual



1. Safety Precautions

1.1. Before using the sensor, please read the instruction manual carefully.
1.2. The sensor is a precision instrument and should be handled with care.
1.3. Do not use the sensor in environments with high temperature, humidity, or vibration.
1.4. The sensor is not waterproof. Do not use it in wet or submerged conditions.
1.5. The sensor is not for use in explosive or flammable atmospheres.

2. Components

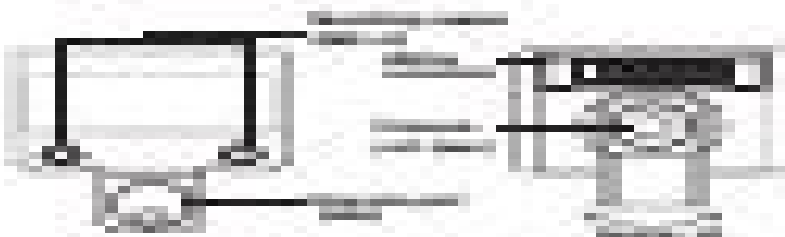


2.1. Sensor Cable

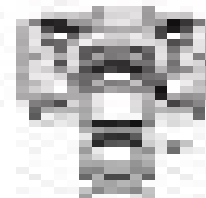
3. Installation



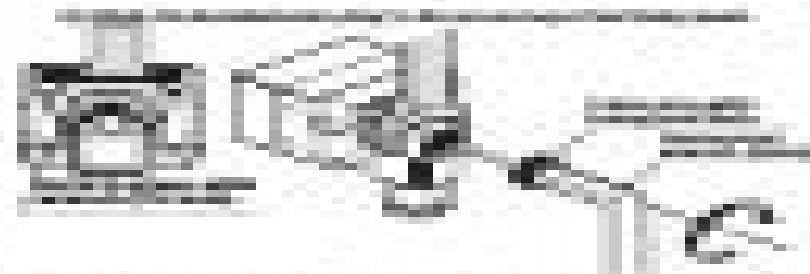
3.1. Connector



2.2. The sensor is a precision instrument and should be handled with care.
2.3. Do not use the sensor in environments with high temperature, humidity, or vibration.
2.4. The sensor is not waterproof. Do not use it in wet or submerged conditions.
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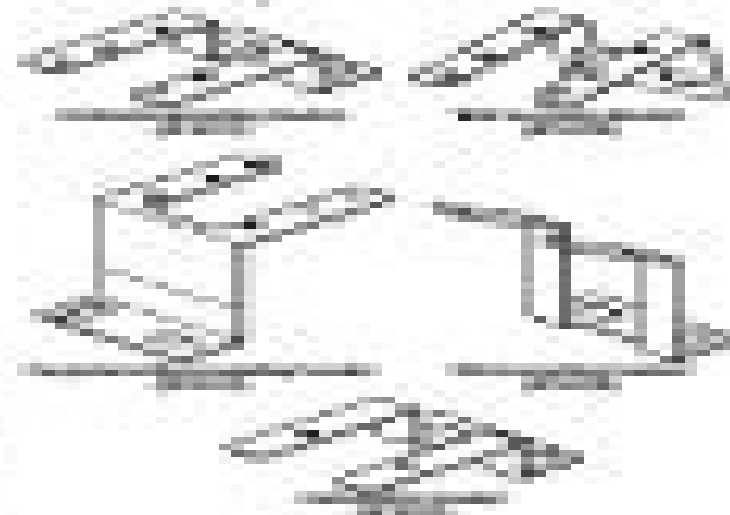


2.5. Connector



2.6. The sensor is a precision instrument and should be handled with care.
2.7. Do not use the sensor in environments with high temperature, humidity, or vibration.
2.8. The sensor is not waterproof. Do not use it in wet or submerged conditions.
2.9. The sensor is not for use in explosive or flammable atmospheres.

4. Operation



Honeywell C30user Guide

Shasha Hu



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web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to each other with single bonds to form a skeleton structure

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web science chemistry chemistry questions and answers molecular modeling purpose in this lab you will employ your skill of drawing lewis dot structures to help you build models of covalently bonded compounds

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web lewis structures a lewis structure is a representation of covalent molecules or polyatomic ions where all the valence

electrons are shown distributed about the bonded atoms as either shared electron pairs bond pairs or unshared electron pairs lone pairs a shared pair of electrons is represented as a short line a single bond

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web molecular models and 3d printing activity lewis dot structures and molecule geometries worksheet answer key 6 10 co 3 2 total number of valence electrons 24 cad engineered 3d sketch model show dipole arrows lewis structure is there a polar bond in this molecule yes or no vsepr shape name trigonal planar bond angles 120

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web jun 8 2022 lewis structure in chemistry lewis structures also called electron dot structures or electron dot diagrams are diagrams that show the bonding between atoms of a molecule and the lone pairs of

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and lone pairs surrounding the central atom and lastly identify the structure geometry