Class: IX

Subject: N. Science(086)

Duration: 3 hours

Maximum Marks: 80

General Instructions:

Read the following instructions very carefully and strictly follow them:

- (i) This question paper comprises 39 questions. All questions are compulsory.
- (ii) This question paper is divided into five sections A, B, C, D and E.
- (iii) Section A Questions No. 1 to 20 are multiple choice questions. Each question carries I mark.
- (iv) Section B Questions No. 21 to 26 are very short answer type questions. Each question carries 2 marks. Answer to these questions should be in the range of 30 to 50 words.
- (v) Section C Questions No. 27 to 33 are short answer type questions. Each question carries 3 marks. Answer to these questions should in the range of 50 to 80 words.
- (vi) Section D Questions No. 34 to 36 are long answer type questions. Each question carries 5 marks. Answer to these questions should be in the range of \$0 to 120 words.
- (vii) Section E Questions No. 37 to 39 are of 3 source-based/case-based units of assessment carrying 4 marks each with sub-parts.
- (viii) There is no overall choice. However, an internal choice has been provided in some sections. Only one of the alternatives has to be attempted in such questions.

			SECTION-A		
Q. No.	Questions The similarity between solid, liquid and gas matter is that, they: a) occupy space b) have mass c) both a and b d) have fixed shape				Marks 1
1					
2	When 20 g of potassium permanganate is dissolved in 1000ml of water, there is no increase in volume. This observation indicates that: a) particles of water have spaces between them into which potassium permanganate particles fit. b) particles of water attract each other. c) particle of potassium permanganate attracts each other. d) particles are continuously moving.				1
3	Which of the follow a) mist	ing is not a Liquid-ga b) clouds	s Aerosol: c) smoke	d) fog	1
-4	The ions present in Na ₂ CO ₃ are:				1

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Table of Contents Grade 9 Science Question Papers Singapore

- 1. Understanding the eBook Grade 9 Science Question Papers Singapore
 - The Rise of Digital Reading Grade 9 Science Question Papers Singapore
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Grade 9 Science Question Papers Singapore
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 9 Science Question Papers Singapore
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Grade 9 Science Question Papers Singapore
 - Personalized Recommendations
 - Grade 9 Science Question Papers Singapore User Reviews and Ratings
 - Grade 9 Science Question Papers Singapore and Bestseller Lists
- 5. Accessing Grade 9 Science Question Papers Singapore Free and Paid eBooks
 - o Grade 9 Science Question Papers Singapore Public Domain eBooks
 - Grade 9 Science Question Papers Singapore eBook Subscription Services
 - o Grade 9 Science Question Papers Singapore Budget-Friendly Options
- 6. Navigating Grade 9 Science Question Papers Singapore eBook Formats
 - o ePub, PDF, MOBI, and More
 - Grade 9 Science Question Papers Singapore Compatibility with Devices
 - Grade 9 Science Question Papers Singapore Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 9 Science Question Papers Singapore
 - Highlighting and Note-Taking Grade 9 Science Question Papers Singapore
 - Interactive Elements Grade 9 Science Question Papers Singapore
- 8. Staying Engaged with Grade 9 Science Question Papers Singapore

- o Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Grade 9 Science Question Papers Singapore
- 9. Balancing eBooks and Physical Books Grade 9 Science Question Papers Singapore
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Grade 9 Science Question Papers Singapore
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 9 Science Question Papers Singapore
 - Setting Reading Goals Grade 9 Science Question Papers Singapore
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 9 Science Question Papers Singapore
 - Fact-Checking eBook Content of Grade 9 Science Question Papers Singapore
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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