

Experiment 5-Worksheet Nomenclature

Write formula or name for each compound:

IA. Binary ionic compounds

| | Name | Formula | Formula | Name |
|---|-------------------|-----------------------|-------------------------|------|
| 1 | sodium oxide | Na_2O | BaS | |
| 2 | calcium nitride | | AlBr_3 | |
| 3 | barium iodide | | Ca_3P_2 | |
| 4 | potassium sulfide | | Mg_3N_2 | |
| 5 | aluminum oxide | | K_2O | |

IB. Transition metals with one oxidation state

| | Name | Formula | Formula | Name |
|---|----------------|---------|-----------------------|------|
| 1 | silver oxide | | ZnCl_2 | |
| 2 | zinc phosphide | | AgI | |
| 3 | zinc sulfide | | Ag_3P | |

IC. Metals more than one oxidation state

| | Name | Formula | Formula | Name |
|---|--------------------|---------|-----------------|------|
| 1 | cupric oxide | | FeBr_3 | |
| 2 | iron (III) sulfide | | CuI | |
| 3 | ferrous nitride | | CuS | |

Formulas And Nomenclature Oxidation Lab

JA Banks



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