

NAMING COMPOUNDS: PRACTICE

Name the following *covalent* compounds:

SO₃ sulfur trioxide

N₂S dinitrogen ^{mono}sulfide

P₂Br₄ diphosphorus tetrabromide

CO carbon monoxide (not ~~monoxide~~)

NO₂ nitrogen dioxide

Formula Writing For Covalent Compounds

**John T. Moore, Chris Hren, Peter J.
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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been

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advanced concepts of chemical science Topics are introduced within the context of familiar experiences whenever possible treated with an appropriate rigor to satisfy the intellect of the learner and reinforced in subsequent discussions of related content The organization and pedagogical features were developed and vetted with feedback from chemistry educators dedicated to the project Dr J SAI CHANDRA Mr SANTOSH RAMCHANDRA KSHIRSAGAR Dr SAMBHAJI MAHIPATI KALE Mr SANDIP PANDURANG GONDAKE Mr SAGAR INDRAJEET SHINDE **Organic Chemistry Volume 1** Roger Macomber, 1996-04-26 This is the first of a two volume set designed for a course focused on the fundamentals of organic chemistry for pre meds and chemistry bioscience students It covers the major aspects of molecular structure followed by an introduction to the techniques of physical and organic chemistry

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