

Assignment #2 – Compound Names and Formulas
Multi-valent ions only

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg₂O
3. gold (III) chloride AuCl₃
4. thallium (III) bromide TlBr₃
5. bismuth (V) oxide Bi₂O₅
6. terbium (III) oxide Tb₂O₃
7. uranium (VI) oxide UO₃
8. protactinium (IV) bromide PaBr₄
9. cerium (III) oxide Ce₂O₃
10. arsenic (V) sulphide As₂S₅

11. manganese (III) oxide Mn₂O₃
12. vanadium (II) bromide VBr₂
13. niobium(V) oxide Nb₂O₅
14. titanium (III) oxide Ti₂O₃
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co₃P₂
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr₂
20. copper (II) bromide CuBr₂

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl₄ Tin (IV) Chloride
2. BiBr₅ Bismuth (V) Bromide
3. PoO₂ Polonium (IV) Oxide
4. PbI₂ Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au₂O₃ Gold (III) Oxide
8. FeCl₂ Iron (II) Chloride

9. PdF₄ Palladium (IV) Fluoride
10. Os₂O₃ Osmium (III) Oxide
11. MoBr₂ Molybdenum (II) Bromide
12. VCl₅ Vanadium (V) Chloride
13. Mn₂O₃ Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np₂O₃ Neptunium (III) Oxide
16. V₂O₅ Vanadium (V) Oxide

Formula Of Ionic Compound Lab Answer Key

Hussin A.Rothana



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U.S. Government Research & Development Reports ,1970 Energy Research Abstracts ,1994-04 Subject Index to Unclassified ASTIA Documents Defense Documentation Center (U.S.),1960 **Scientific and Technical Aerospace Reports** ,1992-11 *The Periodic Table: Nature's Building Blocks* J. Theo Klopprogge,Concepcion P. Ponce,Tom Loomis,2020-11-18 The Periodic Table Nature s Building Blocks An Introduction to the Naturally Occurring Elements Their Origins and Their Uses addresses how minerals and their elements are used where the elements come from in nature and their applications in modern society The book is structured in a logical way using the periodic table as its outline It begins with an introduction of the history of the periodic table and a short introduction to mineralogy Element sections contain their history how they were discovered and a description of the minerals that contain the element Sections conclude with our current use of each element Abundant color photos of some of the most characteristic minerals containing the element accompany the discussion Ideal for students and researchers working in inorganic chemistry minerology and geology this book provides the foundational knowledge needed for successful study and work in this exciting area Describes the link between geology minerals and chemistry to show how chemistry relies on elements from nature Emphasizes the connection between geology mineralogy and daily life showing how minerals contribute to the things we use and in our modern economy Contains abundant color photos of each mineral that bring the periodic table to life *The Software Encyclopedia 2000* Bowker Editorial Staff,2000-05 *Laboratory Text and Notebook for Organic Chemistry* Roy G. Bossert,Wallace Reed Brode,1968 **Chemistry** Frank Jenkins,1992

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