# **How to Find Molecular Mass**

# Molecular mass is the mass of a molecule in daltons (Da or u).

- 1. Start with the molecular formula.
- Look up the atomic masses of the elements.
- Add up the atomic masses according to the molecular formula. To account for the correct number of atoms, multiply the atomic masses by their subscripts.

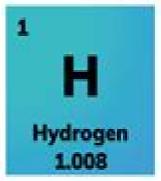


# Example

Find the molecular mass of methane, CH<sub>4</sub>.

Atomic mass of C (carbon) is 12.011. Atomic mass of H (hydrogen) is 1.008.

Carbon 12.011



Molecular mass =  $12.011 + (4 \times 1.008)$ 

Molecular mass = 12.011 + 4.032

Molecular mass = 16.043

# **Formula And Molar Mass Answers**

Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas

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