

A-Level Chemistry - Year 2

OCR Chemistry A - Paper 1 June - 2019

16 Sir Humphrey Davy discovered several elements including sodium, potassium, magnesium, calcium and strontium.

(a) Explain which block in the Periodic Table sodium and magnesium belong to.

(1)

(b) A sample of magnesium, $A_r = 24.305$, is found to consist of three isotopes. The accurate relative isotopic masses and % abundances of two of the isotopes are shown in the table.

| Isotope | Relative isotopic mass | % abundance |
|------------------|------------------------|-------------|
| ^{24}Mg | 23.985 | 78.99% |
| ^{26}Mg | 24.986 | 10.00% |

Determine the relative isotopic mass of the third isotope of magnesium in the sample.

Give your answer to 5 significant figures.

relative isotopic mass = _____ (2)

Question - 16

OCR

International Centre for

Tuesday 4 June 2019 – Afternoon

A Level Chemistry A

H432091 Periodic table, elements and physical chemistry

Time allowed: 2 hours 15 minutes

We need from:

1. An A-level Chemistry A (or equivalent) qualification

We also need:

1. A minimum of 1000 hours of study



Please write clearly in black ink. Do not write in the brackets.

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Candidate number

First name(s)

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INSTRUCTIONS

- Use black ink. You may use an HB pencil for graphs and diagrams.
- Answer all the questions.
- Where appropriate, your answers should be supported with writing. Marks may be given for a correct method even if the answer is incorrect.
- Write your answer to each question in the space provided. If additional space is required, use the back paper at the end of this booklet. The question numbers must be clearly shown.

INFORMATION

- The total mark for this paper is 100.
- The marks for each question are shown in brackets ().
- Quality of extended responses will be assessed in questions marked with an asterisk (*).
- This document consists of 32 pages.

F321 Chemistry June 2014 Paper

Didier Musso



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