

- 20) What is **not** a typical characteristic of acids?
- A** They react with alkalis producing water.
  - B** They react with **all** metals producing hydrogen.
  - C** They react with carbonates producing carbon dioxide.
  - D** They turn blue litmus paper red.

- 21) Zinc sulfate is made by reacting an excess of zinc oxide with dilute sulfuric acid. The excess zinc oxide is then removed from the solution. Which process is used to obtain solid zinc sulfate from the solution?
- A** crystallisation
  - B** dissolving
  - C** filtration
  - D** fractional distillation

- 23) Three solids, P, Q and R, all react with dilute sulfuric acid to produce zinc sulfate. P and R produce gases during the reaction. The gas produced when P reacts will not burn. The gas produced when R reacts will burn. What are P, Q and R?

	P	Q	R
<b>A</b>	zinc	zinc hydroxide	zinc carbonate
<b>B</b>	zinc carbonate	zinc	zinc oxide
<b>C</b>	zinc oxide	zinc hydroxide	zinc
<b>D</b>	zinc oxide	zinc carbonate	zinc

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