#### Worksheet: Writing and Balancing Chemical Reactions

- Balance the following equations and indicate the type of reaction as formation, decomposition, single replacement, double replacement, hydrocarbon combustion, or other.
- a.  $C_0 \hookrightarrow C_0 \hookrightarrow C$ **b.** \_\_\_\_  $H_2O_{(0)} \rightarrow$ \_\_\_\_  $H_{2((0))} +$ \_\_\_\_  $O_{2((0))}$ c. \_\_\_\_ Fe<sub>(i)</sub> + \_\_\_\_ H<sub>2</sub>O<sub>(g)</sub>  $\rightarrow$  \_\_\_\_ H<sub>2(g)</sub> + \_\_\_\_ Fe<sub>3</sub>O<sub>4(g)</sub> d. \_\_\_\_ AsCl<sub>3 (ee)</sub> + \_\_\_\_ H<sub>2</sub>S (ee)  $\rightarrow$  \_\_\_\_ As<sub>2</sub>S<sub>3 (e)</sub> + \_\_\_\_ HCl (ee) e. \_\_\_\_ CuSO<sub>4</sub> • 5  $H_2O_{(0)} \rightarrow$  \_\_\_\_ CuSO<sub>4(0)</sub> + \_\_\_\_  $H_2O_{(0)}$ f. \_\_\_\_ Fe<sub>2</sub>O<sub>3  $\omega$ 0 + \_\_\_\_ H<sub>2</sub> $\omega$ 0  $\rightarrow$  \_\_\_\_ Fe $\omega$ 0 + \_\_\_\_ H<sub>2</sub>O $\omega$ 0</sub> g.  $CaCO_{3 (n)} \rightarrow CaO_{(n)} + CO_{2 (n)}$ h. \_\_\_\_ Fe  $_{(0)}$  + \_\_\_\_ S<sub>8 (0)</sub>  $\rightarrow$  \_\_\_\_ FeS  $_{(0)}$ i. \_\_\_\_  $H_2S_{(aq)} +$  \_\_\_\_  $KOH_{(aq)} \rightarrow$  \_\_\_\_  $H_2O_{(0)} +$  \_\_\_\_  $K_2S_{(aq)}$ j. \_\_\_\_ NaCl $_{(0)} \rightarrow$  \_\_\_\_ Na $_{(0)} +$  \_\_\_\_ Cl $_{2(0)}$ k. \_\_\_\_  $Al_{(0)}$  + \_\_\_\_  $H_2SO_{4(aq)}$   $\rightarrow$  \_\_\_\_  $H_{2(q)}$  + \_\_\_\_  $Al_2(SO_4)_{3(aq)}$ 1. \_\_\_\_\_  $H_3PO_4$  (oat) + \_\_\_\_\_  $NH_4OH$  (oat)  $\rightarrow$  \_\_\_\_\_  $H_2O$  (t) + \_\_\_\_\_ ( $NH_4$ )  $_3PO_4$  (oat) m. \_\_\_\_  $C_3H_{8(g)} +$ \_\_\_\_  $O_{2(g)} \rightarrow$ \_\_\_\_  $CO_{2(g)} +$ \_\_\_\_  $H_2O_{(f)}$ o. \_\_\_\_  $CH_{4(g)} + ____ O_{2(g)} \rightarrow ____ CO_{2(g)} + ____ H_2O_{4h}$ p. \_\_\_\_  $K_2SO_{4 (aq)} +$ \_\_\_\_  $BaCl_{2 (aq)} \rightarrow$ \_\_\_  $KCl_{(aq)} +$ \_\_\_  $BaSO_{4 (a)}$ q.  $C_5H_{12(0)} + C_2G_{00} \rightarrow CO_{2(6)} + H_{2O(6)}$ r. \_\_\_\_ Ca(OH)<sub>2 (aq)</sub> + \_\_\_\_ NH<sub>4</sub>Cl<sub>(aq)</sub>  $\rightarrow$  \_\_\_\_ NH<sub>4</sub>OH<sub>(aq)</sub> + \_\_\_\_ CaCl<sub>2 (aq)</sub> s.  $\underline{\hspace{1cm}} V_2O_5\omega_1 + \underline{\hspace{1cm}} Ca\omega_2 \rightarrow \underline{\hspace{1cm}} CaO\omega_1 + \underline{\hspace{1cm}} V\omega_2$ t.  $Na_{co} + Znl_{2coo} \rightarrow Nal_{coo} + Zn_{co}$ u. \_\_\_\_  $C_2H_6O_3 + ___ O_2 (y) \rightarrow ___ CO_2 (y) + ___ H_2O (y)$ w. \_\_\_\_  $Fe_2O_{3(x)} + ___ H_{2(x)} \rightarrow ___ Fe_{(x)} + ___ H_2O_{(0)}$ x.  $C_{15}H_{30,00} + C_{2,(e)} \rightarrow C_{2,(e)} + H_{2}O_{60}$ y. \_\_\_\_  $BN_{(0)} + ____ F_{2(0)} \rightarrow ____ BF_{3(0)} + ____ N_{2(0)}$ z.  $C_{12}H_{26(0)} + C_{2(0)} \rightarrow C_{2(0)} + H_{2}O_{(0)}$
- Predict the product(s) along with the states, indicate the type of reaction, and balance the following chemical reactions.
- a. A solution of lead (II) nitrate is mixed with a solution of sodium iodide.
- b. Solid zinc sulfide reacts with oxygen in the air.
- Liquid butane (C<sub>4</sub>H<sub>10 (f)</sub>) is used as a fuel to ignite a lighter.
- Barium hydroxide solution is neutralized by adding hydrochloric acid (HCl (app)).
- Copper metal is placed in a solution of silver nitrate.
- f. Sulfur burns in oxygen to make sulfur dioxide gas.
- g. A solution of aluminum sulfate is mixed with a solution of calcium hydroxide.
- Zinc metal is placed in sulfuric acid (H<sub>2</sub>SO<sub>4 (aq)</sub>).
- Aluminum powder is placed in a container filled with chlorine gas.
- Sucrose undergoes cellular respiration.

# <u>Identifying And Balancing Chemical Equations Answers</u> <a href="#">Chapter 8</a>

**Ralph Thomas Myers** 

# **Identifying And Balancing Chemical Equations Answers Chapter 8:**

**Chemistry** John Olmsted, Gregory M. Williams, 1997 Textbook outling concepts of molecular science **Basic Concepts** of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance **Foundations** of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence. The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity Holt Chemistry Ralph Thomas Myers, 2004 Introduction to Chemical Principles Edward I. Peters, 1986 Design of Simple and Robust Process Plants J. L. A. Koolen, 2001-10-15 The approaches to design process plants described in this book lead to process designs which require 30 40% less capital than usual The book is unique since it is the first comprehensive work addressing both the total process design and operational approach Technological developments during the last decade made the design of really competitive processes possible Mechanical developments have resulted in reliable and robust equipment Process developments have created opportunities to minimize the amount of equipment furthermore different logistic approaches integration of process functionality and intensification of the unit operations are possible Computer and control technology allows remote control operation and first pass prime production In this work design philosophies are discussed and their implementation is shown as a structured approach for planned and existing plants Numerous examples are presented to illustrate what simple design can create The work is intended for experienced engineers and managers involved in process design control design and

operation but is also interesting for students Project engineers and managers have to apply these new approaches to achieve competitive processes A process plant should meet the simplicity and robustness of a household refrigerator This book has been written to allow to achieve this aim Chairman of the Judges Award from IChemE 2003 **Prentice Hall Chemistry Solutions Guide for Introductory Chemistry** James F. Hall,1996 Oswaal NCERT Textbook Solution Class 10 Science & Mathematics | Set of 2 Books | For Latest Exam Oswaal Editorial Board, 2024-03-30 Description of the product 100 % Updated as per latest textbook issued by NCERT Crisp Revision with Concept wise Revision Notes Mind Maps and Mnemonics Visual Learning Aids with theoretical concepts and concept videos Complete Question Coverage with **Chemical Interactions** .2005 all Intext questions and Exercise questions Fully solved Educational Assessment of Students Anthony J. Nitko, 2004 A highly respected book in its field The Educational Assessment of Students 4 e is the most thorough discussion of traditional and alternative assessments of any text explaining giving practical real world examples discussing pros and cons and showing how to construct virtually all of the choices teachers can make in classroom assessment Theories and research findings abound the author examines why when and how teachers should use assessment in the classroom Topics include the bases for assessment in the classroom crafting and using classroom assessments and interpreting and using standardized tests With its extensive and valuable appendix set this book is a must have desk reference for teachers and others involved in the field of education Stormwater Effects Handbook G. Allen Burton Jr., Robert Pitt, 2001-08-29 A stand alone working document Stormwater Effects Handbook A Toolbox for Watershed Managers Scientists and Engineers assists scientists and regulators in determining when stormwater runoff causes adverse effects in receiving waters This complicated task requires an integrated assessment approach that focuses on sampling **Current Engineering Practice** ,1978 **Fundamentals of Chemical Reaction** before during and aft **Engineering** Mark E. Davis, Robert J. Davis, 2013-05-27 Appropriate for a one semester undergraduate or first year graduate course this text introduces the quantitative treatment of chemical reaction engineering It covers both homogeneous and heterogeneous reacting systems and examines chemical reaction engineering as well as chemical reactor engineering Each chapter contains numerous worked out problems and real world vignettes involving commercial applications a feature widely praised by reviewers and teachers 2003 edition **Ebook: Chemistry: The Molecular Nature of Matter and Change** Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change Physical Science - Chemistry Split with Online Learning Center Password Card (Chapters 1 And 8 - 13) Bill W. Tillery, Tillery Bill, 2004-01-09 Oswal -Gurukul Chemistry Most Likely Question Bank: ICSE Class 10 For 2023 Exam Oswal - Gurukul, 2022-05-14 Enclosure Fire Dynamics, Second Edition Björn Karlsson, James G. Quintiere, 2022-06-27 Enclosure Fire Dynamics Second Edition explores the science of enclosure fires and how they cause changes in the environment of a building on fire The authors discuss mechanisms controlling enclosure fires and how to develop analytical relationships useful in designing

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